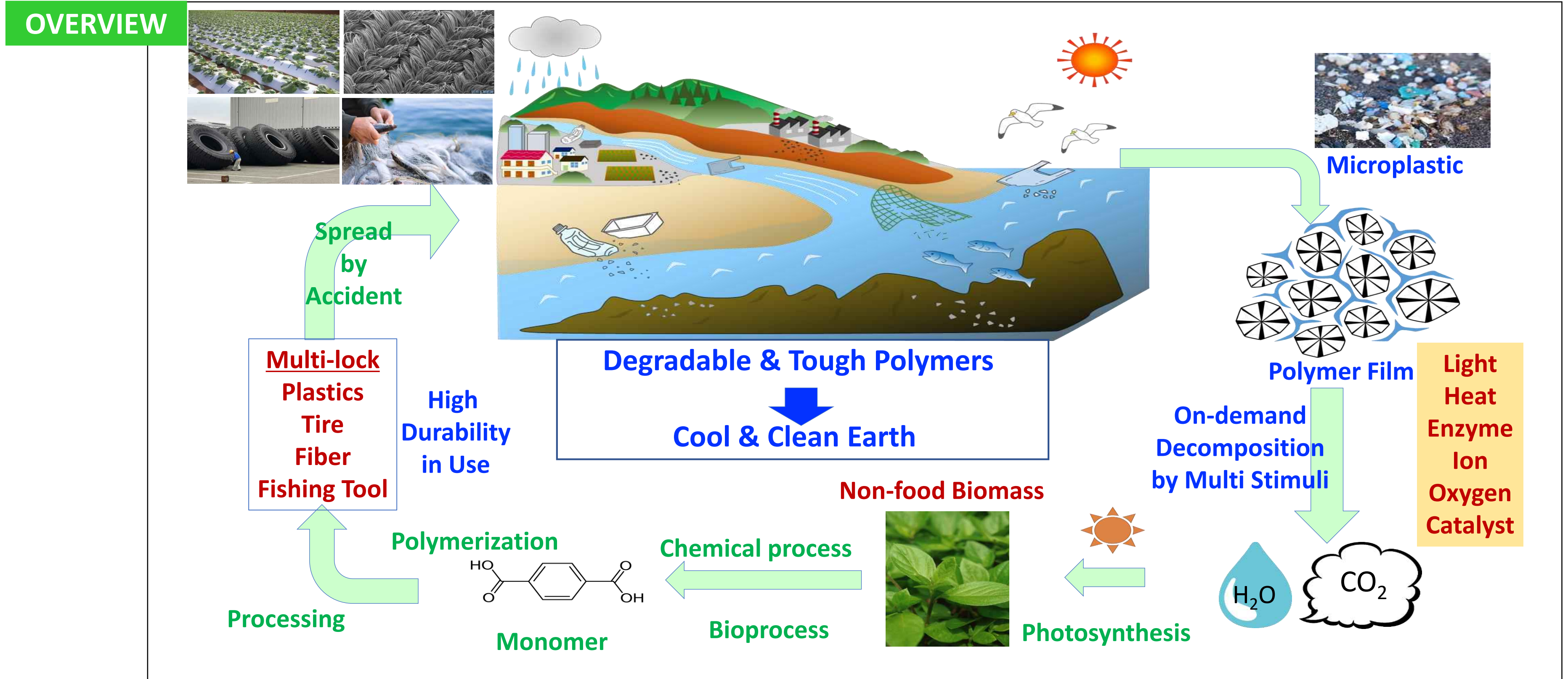


Organization: University of Tokyo

Contact: Kohzo Ito (kohzo@edu.k.u-tokyo.ac.jp), Naoki Kato (naokikato@g.ecc.u-tokyo.ac.jp)



For seawater decomposition of **difficult-to-collect plastics, tire wear debris, fiber scraps, and fishing gear**, we have introduced a multi-lock mechanism that achieves both degradability and durability, and realizes on-demand disassembly. In addition, by using non-edible biomass as a raw material, CO2 reduction can be achieved at the same time.



Organization

NEDO
Materials Technology and Nanotechnology Department
Bioeconomy Promotion Division

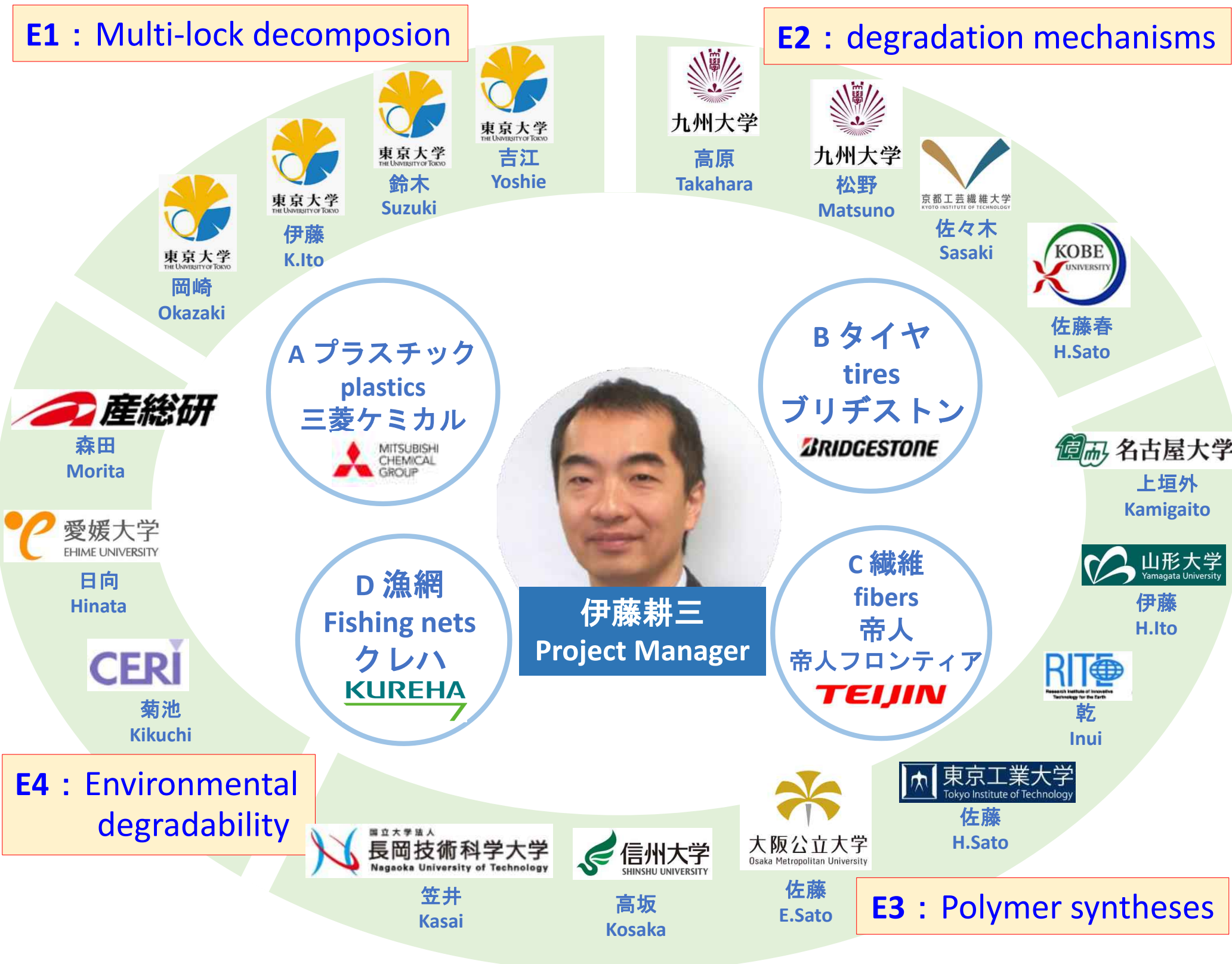
ITO P J
project manager
Kohzo ITO
PROJECT OFFICE

Entrusted organizations : 12 inst.

Re-entrusted organizations: 7inst.

Company : 5 inst.
Academia : 14 inst.
(18Gr)

Entrusted organization	Leader
2 Mitsubishi Chemical Corporation	Atsushi KUSUNO
3 Bridgestone	Satoshi HAMATANI
4 TEIJIN	Tomoyoshi YAMAMOTO
TEIJIN FRONTIER	Kazuhiro MORISHIMA
5 KUREHA	Takashi MASAKI
The University of Tokyo	Kohzo ITO Susumu OKAZAKI Naoko YOSHIE Kosuke SUZUKI
1 Kyoto Institute of Technology	Sono SASAKI
Kobe University	Harumi SATO
Osaka City University	Eriko SATO
SHINSHU UNIVERSITY	Yasuhiro KOHSAKA
Nagaoka University of Technology	Daisuke KASAI
CERI	Takako KIKUCHI
6 KYUSHU UNIVERSITY	Atsushi TAKAHARA Hisao MATSUNO
7 Nagoya University	Masami KAMIGAITO
8 Yamagata University	Hiroshi ITO
9 RITE	Msayuki INUI
10 AIST	Hiroshi MORITA
11 EHIME UNIVERSITY	Hirofumi HINATA
12 Tokyo Institute of Technology	Kotaro SATOH



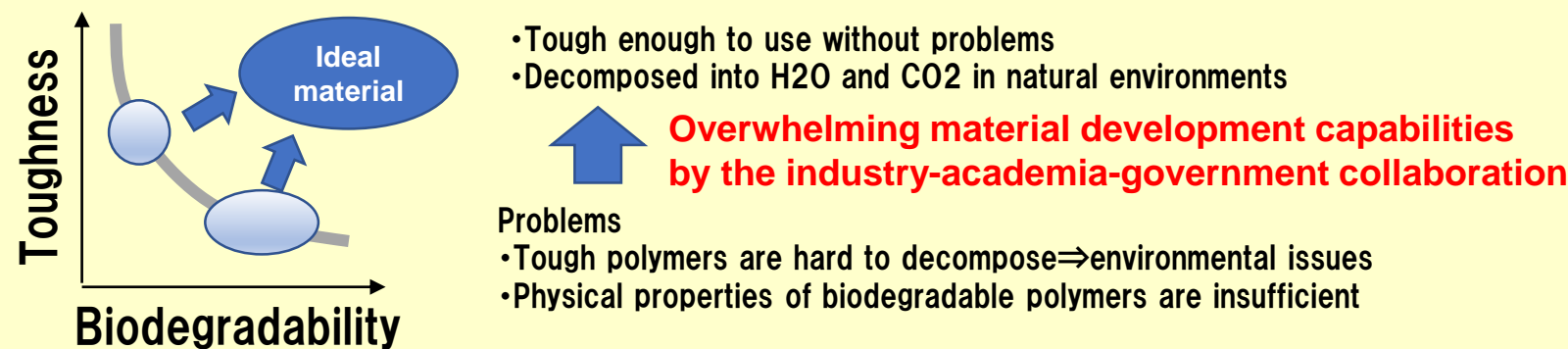
Common issues	Targets	Member
E1+E3 Multi-lock decomposition mechanism (switch function)	Develop a multi-lock degradation mechanism for model resins and elastomers that can be degraded on demand by multiple stimuli expected in the marine.	UT, NU, RITE, TIT, AIST, OCU, SU, NUT
E2 Elucidation of degradation mechanisms	Elucidate the degradation mechanisms of model resins and elastomers in natural environments, including the ocean.	Kyushu U, KIT, Kobe U, AIST, CERI
E3-1 Polymer syntheses from non-food Biomass	Monomers made from non-food biomass will be synthesized using enzymes and organic synthesis.	NU, RITE, TIT, SU
E3-2 Improving the Durability and Toughness	Improve the durability and toughness of environmentally degradable polymers.	YU, Kyushu U, UT, NU, AIST
E4 Evaluation of environmental degradability	The dynamic analysis of plastic wastes in the ocean and the development of a fast decomposition evaluation method.	EU, CERI
E5 Marine safety of oligomers	Synthesize oligomers and evaluate their marine degradability and safety.	Kyushu U, NU, TIT, SU, CERI

1. Concept and objectives



[Objectives] The purpose is to develop a bioplastic that **incorporates a multi-locking mechanism** in aliphatic polyesters produced from inedible resources and that **quickly biodegrades in seawater** after being unlocked by multiple external stimuli. We also aim to toughen biodegradable plastics while maintaining good biodegradability by introducing dynamic cross-linking or supramolecules and optimizing of higher-order structures. In this work, we will investigate the introduction of multi-locking mechanism and toughening of PBS (polybutylene succinate) resin.

[Concept] Moonshot program led by the Cabinet Office
Achieve both high toughness and high biodegradability



3. Plan (FY2020-FY2029)

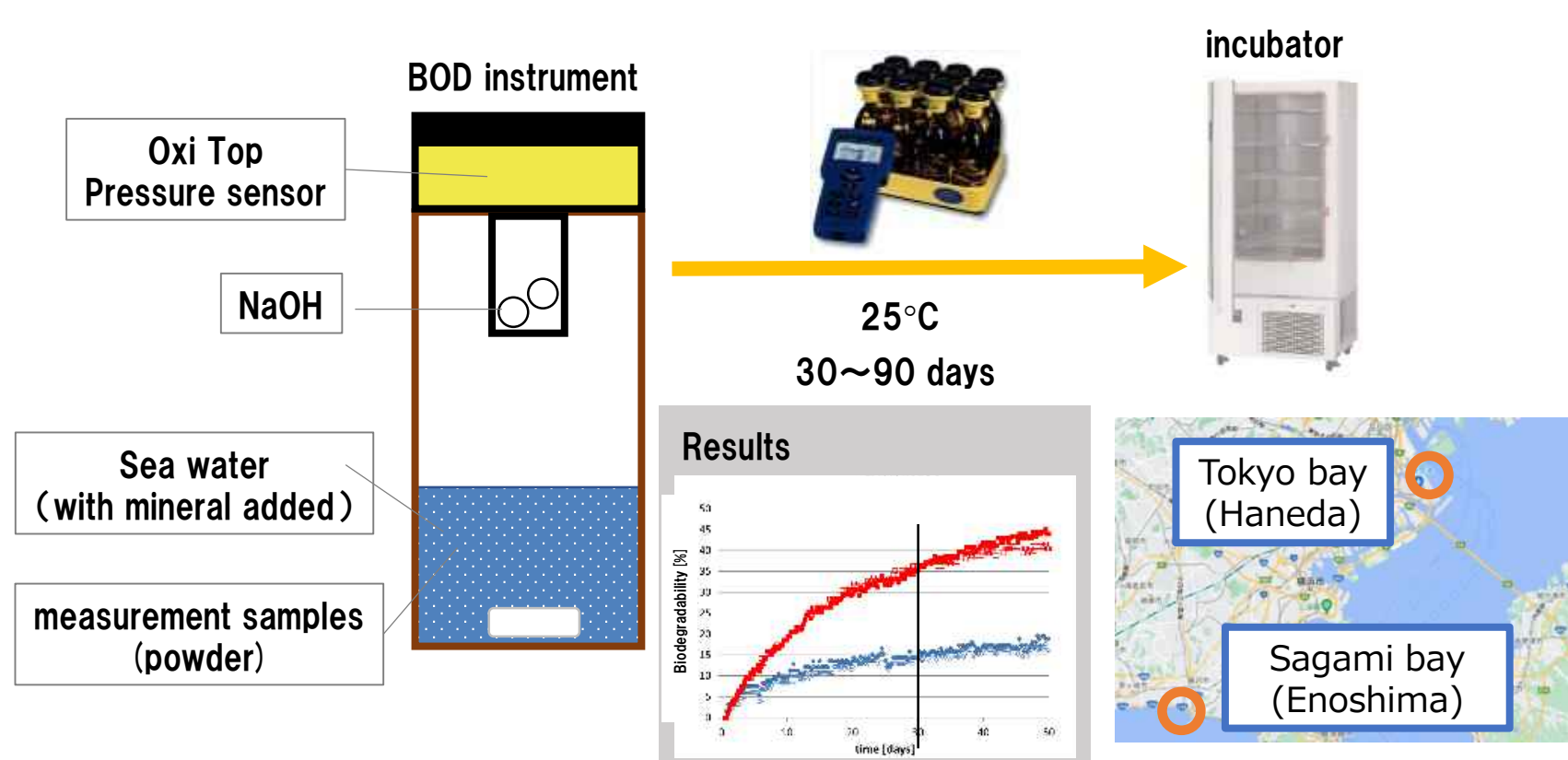


Fiscal Year	20	21	22	23	24	25	26	27	28	29
Intermediate/Final target			▲		▲			▲		▲
Multi-locking mechanism		concept proof				optimization		Scale-up		
Toughening (Improving tear strength)		Establishing the methodology				Optimization, scale-up (Kneading)				
Production						Scale-up				
Material properties		Marine biodegradability evaluation								
		Evaluation of physical, thermal and practical properties								

5. Evaluation of marine biodegradability by BOD test



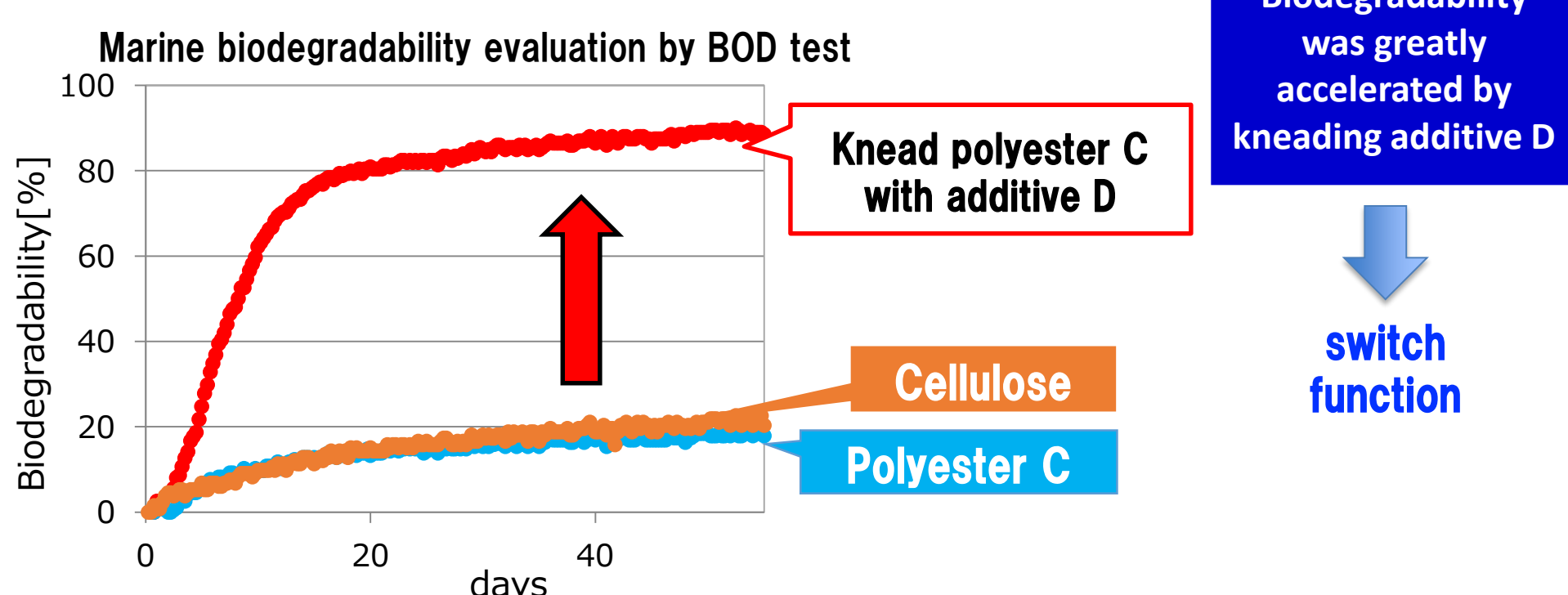
BOD: Biochemical Oxygen Demand [mg/L]
 Biodegradation: consume O₂, produce CO₂
 BOD test: measuring O₂ consumption (NaOH absorbs CO₂) → calculate degree of biodegradability



7. Results of marine biodegradability improvement study



Additives were investigated to improve marine biodegradability of copolymerized polyesters.



• Coexistence of "certain additives" with marine biodegradable resins → Improved marine biodegradability

2. Targets



- FY2022 Intermediate Target: **Proof of the multi-locking concept**
 - Degradation rate is more than 3 times higher for multiple external stimuli than for a single external stimulus.
- FY2024 Intermediate Target: **Achieve both high toughness and multi-locking mechanisms**
 - Degradation rate is more than 10 times higher for multiple external stimuli than for a single external stimulus.
 - 5 times higher tear strength than existing aliphatic polyesters
- FY2027 Intermediate Target: **Demonstration of the Bench-scale production**
 - Can be manufactured in scales of 20 kg or more
- FY2029 **Final Target**: Achieve the followings with scaled-up products
 - Marine biodegradation after unlocked :40% biodegradability in sea water (25°C) in 30 days.
 - **Tear strength: More than 10 times** that of existing biopolymers
 - Polymer production on a scale larger than bench scale

4. Academia study contents

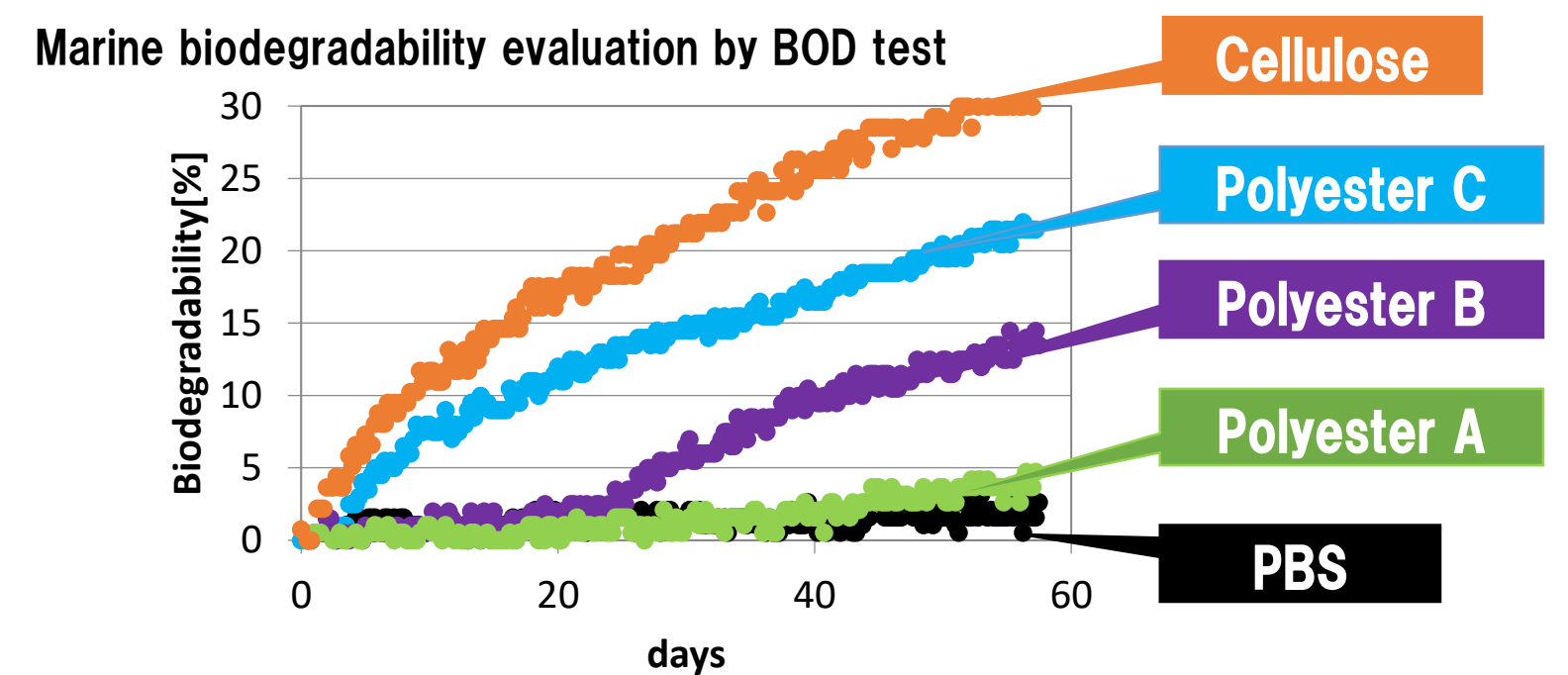


Study items	Targets (in the end of FY2022)		PIC
	Primary	secondary	
Biodegradability	Relationship between biodegradability and hydrogen bonding	Proposal for a primary structure with excellent marine biodegradation	Kobe Univ. Prof. Sato
	Effect of higher order (interfacial) structure (Biofilm, bacteria, Interaction with enzyme)	Proposal for a primary structure with excellent marine biodegradation	Kyushu Univ. Prof. Takahara Prof. Matsuno
Biodegradability /Toughness	Improve marine biodegradability and toughness	Proposal formulation with excellent marine biodegradation	Tokyo Univ. Prof. Ito
Toughness	Relationship between crystal structure and tear strength	Compounding with superior toughening and process study	Yamagata Univ. Prof. Ito
	Crystal structure changes before and after tearing (in situ)	Establishing analysis method	Kyoto institute of technology Prof. Sasaki

6. Results of marine biodegradability improvement study



Copolymer composition study to improve marine biodegradability of PBS



• New comonomers are added through copolymerization
 → Improved marine biodegradability

8. Summary and Future Plans



●FY2022 Results

- The crystal structure change before and after in situ tearing was clearly observed. (Prof. Sasaki, Kyoto Institute of Technology)
- Retardation analysis of the tearing tip confirmed the difference in behavior between polymers. (Prof. Ito, Yamagata Univ.)
- Improvement of toughness and biodegradability was confirmed by adding PR (polyrotaxane) to PBSA (polybutylene succinate adipate). (Tokyo Univ., Prof. Ito)
- Changes in crystal structure and hydrogen bonding state of copolymerized biodegradable resins were confirmed (Prof. Sato, Kobe Univ.)
- The higher-order structural change by weathering test and the promotion of biofilm formation by weathering test degradation were confirmed. (Prof. Takahara, Kyushu Univ.)
- Surface structure change by hydrolytic enzyme was confirmed. (Prof. Matsuno, Kyushu Univ.)
- Improvement of biodegradability was confirmed by examining the copolymerization composition of PBS and additives. (Mitsubishi Chemical Corporation)

● Achievement level against target: Progress as planned

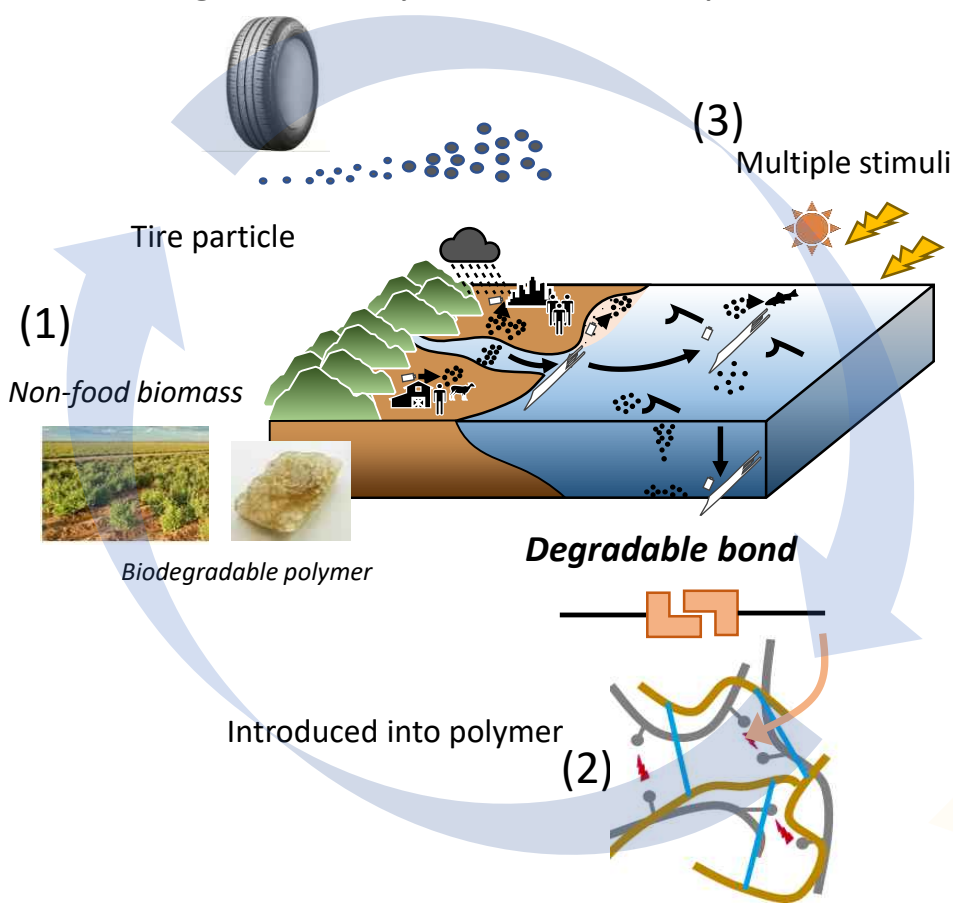
●Future Plans

- (1) Degradability control (elucidation of degradation mechanism and introduction of trigger mechanism)
- (2) Improvement of tearing strength (elucidation and improvement of tearing)

Research outline of this project

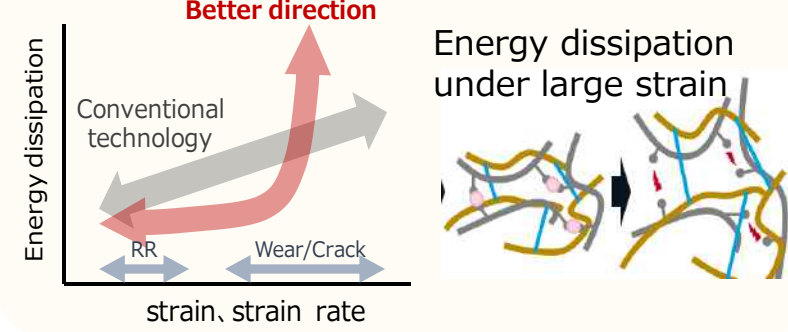
In recent years, there has been growing concern about the influence of tire wear particle on marine as microplastics. While its substantial contribution to the environment is still debatable, technological development is desired from a view of environmental pollution/circulation of resources. In this study, we aim to solve these issues by developing non-food biomasses based multi-lock tough polymer which can be decomposed by multiple stimuli. Combined with the toughness technology by energy dissipation proposed in IMPACT project (2014-2019), the developed tough polymer is applied to tire tread, and it demonstrates toughness by effective energy dissipation in use and quickly decomposes by multiple stimuli (microorganism and combination of light, heat, oxygen, etc.) after use in the state of wear particle.

—Biodegradation cycle of tire wear particle—



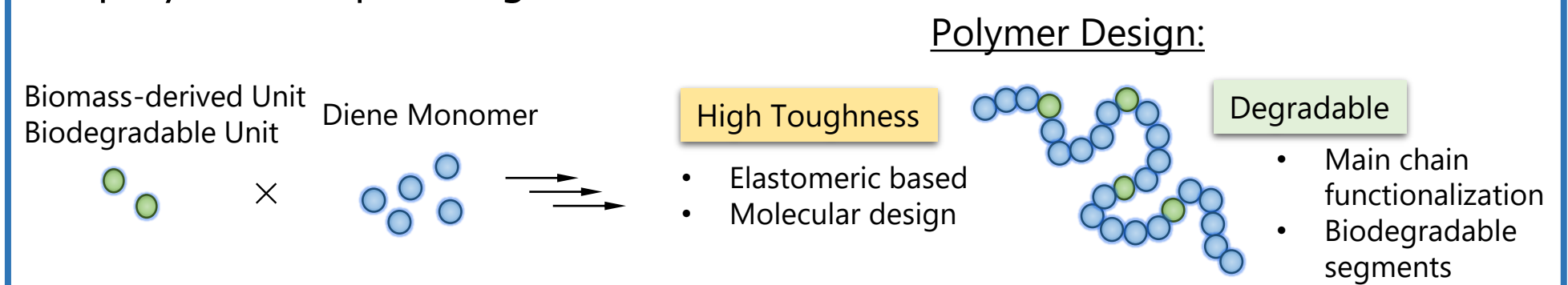
- (1) development of non-food biomasses based biopolymer synthesis
- (2) development of multi-lock degradability technology
- (3) development of degradability evaluation method/degradability
- (4) compound design for highly balancing degradability and toughness by energy dissipation

(4) —Toughness by energy dissipation—



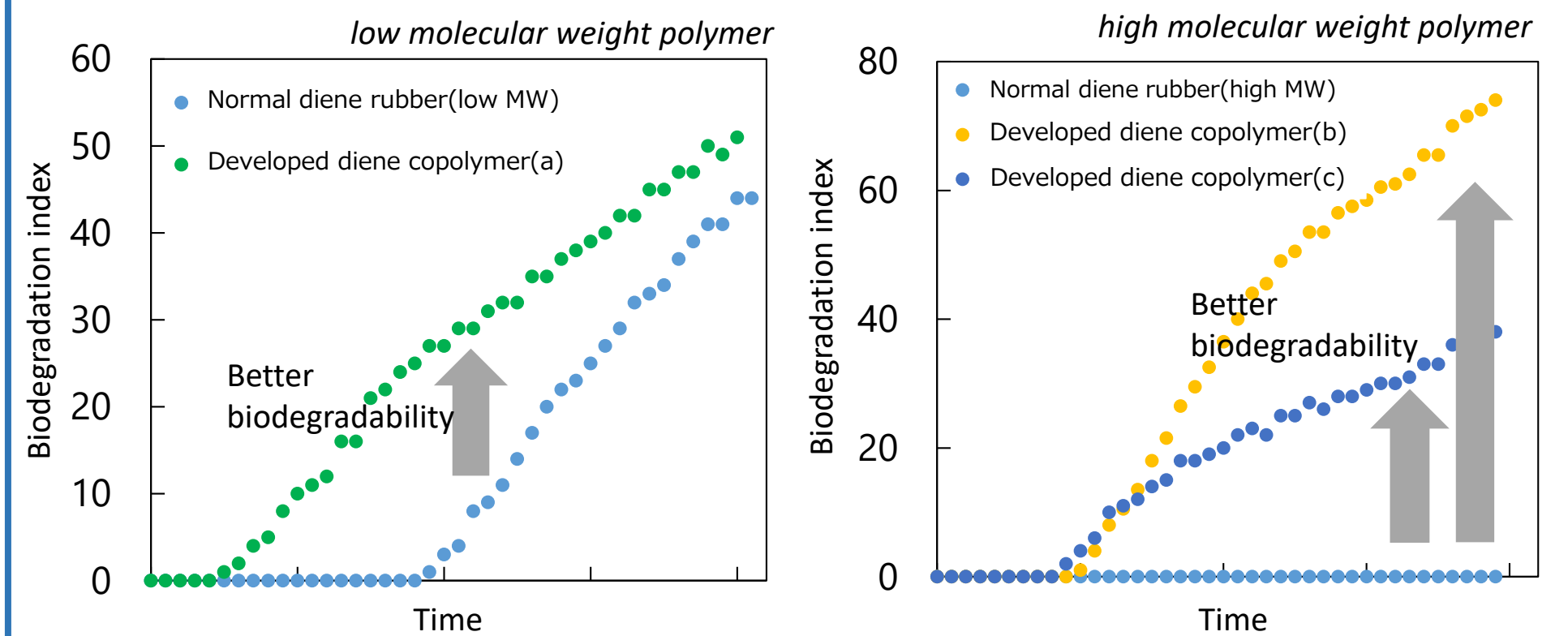
Biodegradation technology by introducing degradable unit

Biomass-derived degradable unit was introduced to diene rubber copolymer cooperating with academia.



We have succeeded in synthesizing diene rubber copolymers which was introduced several biomass-derived units/degradable units. It was verified that these copolymers can improved the biodegradation rate.

—Marine biodegradation test result—



Improved biodegradability compared with the same molecular weight

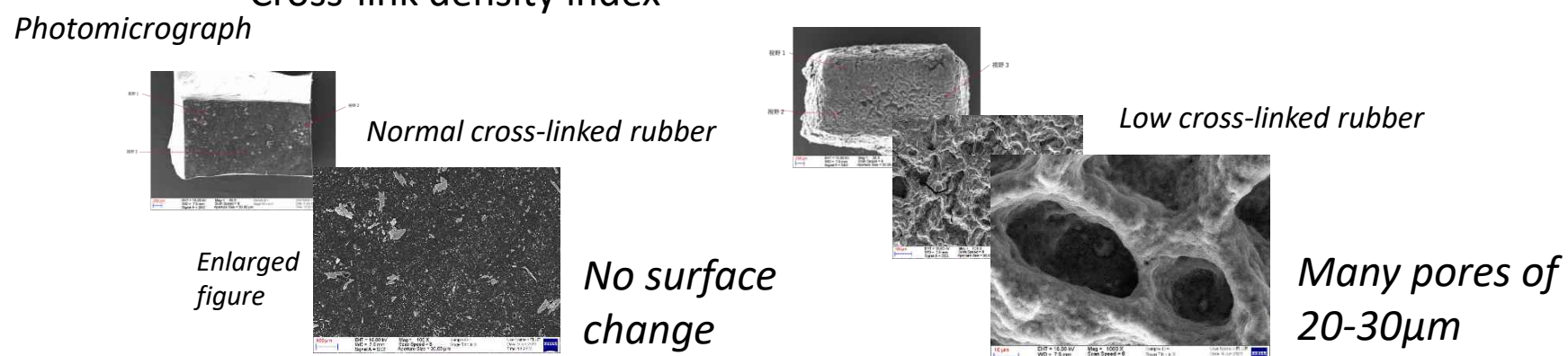
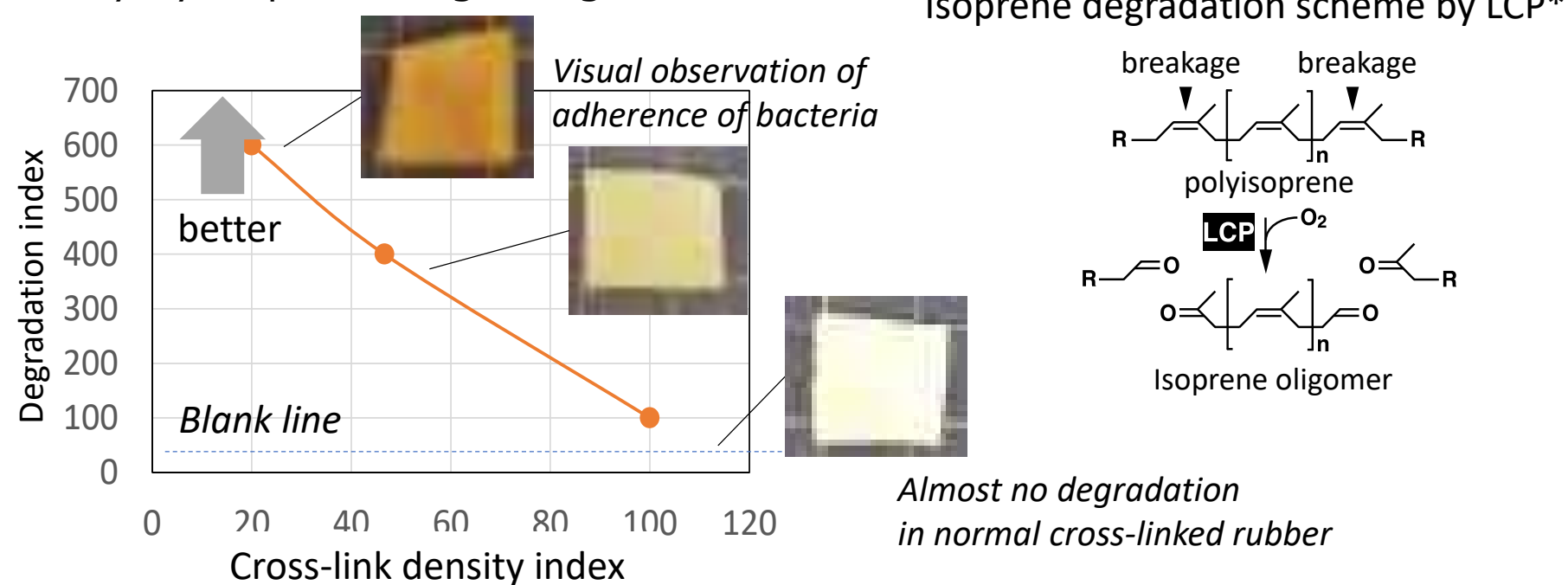
Very effective for high molecular weight polymer

Biodegradability improved by more than 10 times with degradable unit introduction rubber

Degradability analysis of rubber

Rubber biodegradation mechanism is carried out cooperating with academia. The effect of cross-link on rubber degradation behavior was analyzed using the bacteria that has ability to decompose cross-linked rubber. Contributions of cross-link density and molecular weight to degradation behavior have been clarified.

—Degradation test result of rubber with different cross-link density by isoprene-degrading bacteria—

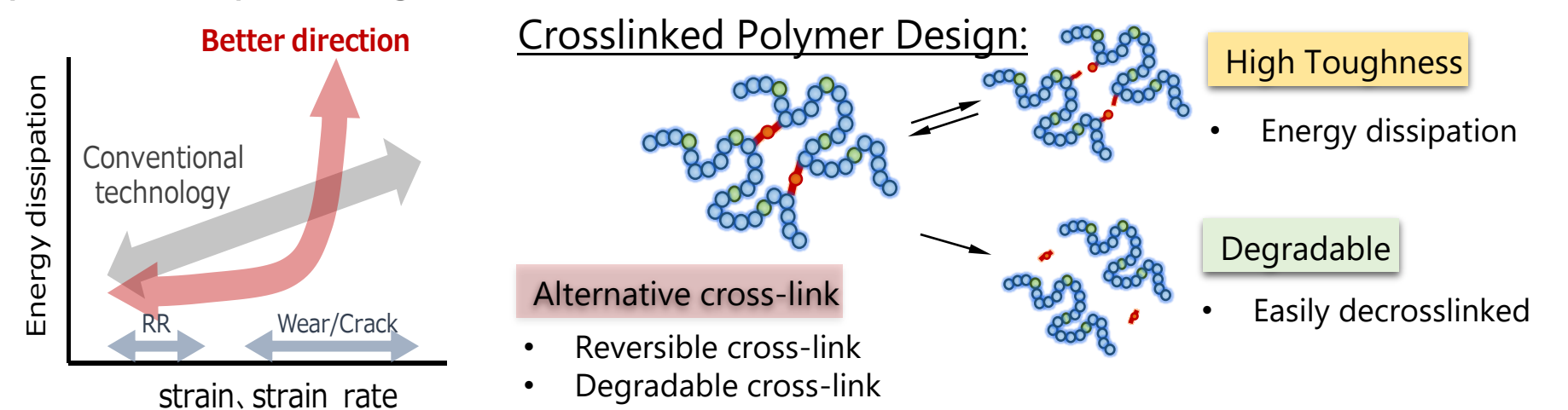


Natural rubber and synthetic isoprene rubber showed same degradation trend in polymer main chain. No degradation was observed in conventional cross-link density. In our studies, we have found that the degradation rate is strongly associated with cross link density and molecular weight. With decreasing cross link density and molecular weight, degradation rate increases.

*D.Kasai: Kagaku to Seibutsu 58(2) 77 (2020)

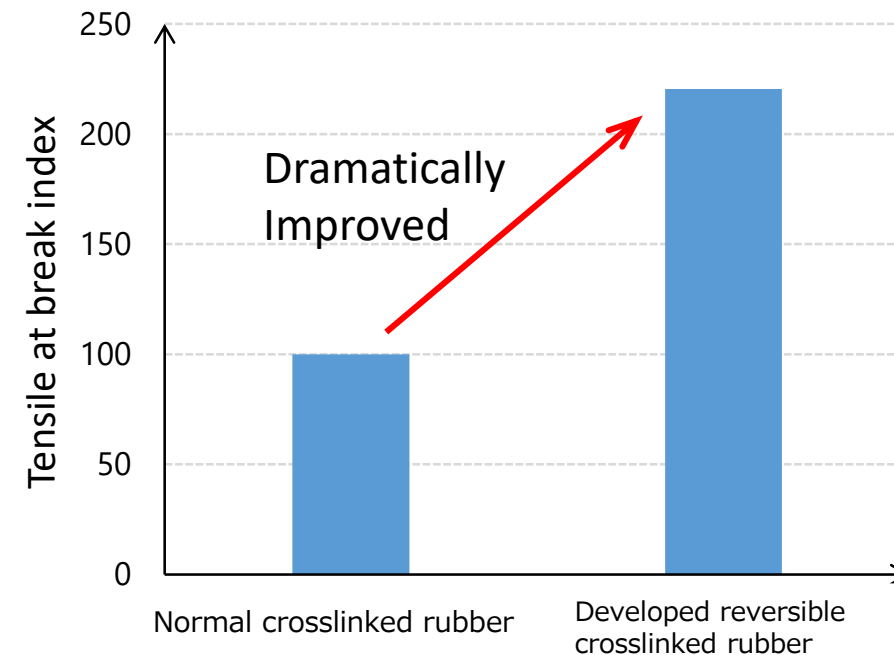
Technology for balancing biodegradation and toughness by reversible cross-link

With extending reversible bond technology that strengthen rubber by effective energy dissipation, we newly designed degradable cross link system cooperating with academia.

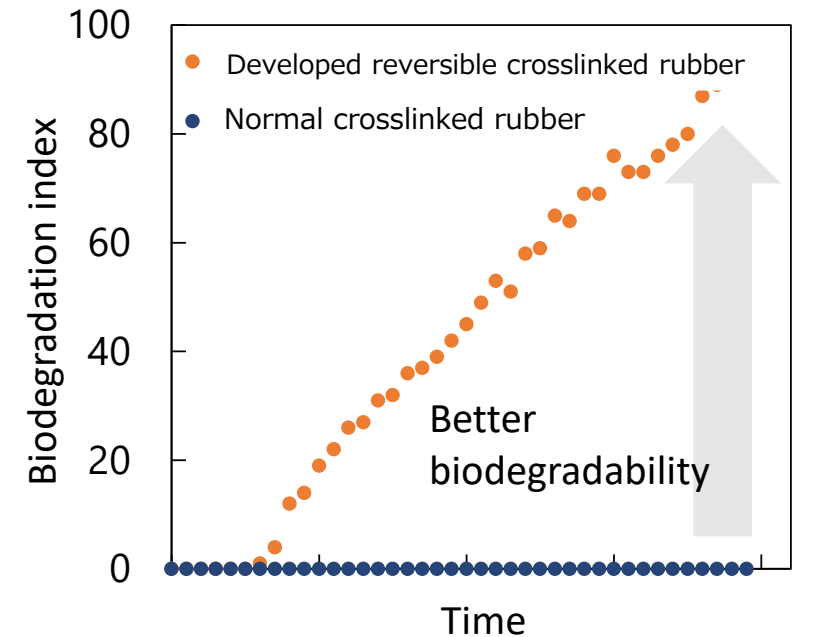


We have succeeded in improving the toughness of rubber by energy dissipation through reversible and degradable cross-links. In addition, we developed diene rubber copolymers with reversible and degradable cross-links and verified whether the toughness and biodegradation rate of these copolymers are both improved.

—Physical property test result—



—Marine biodegradation test result—

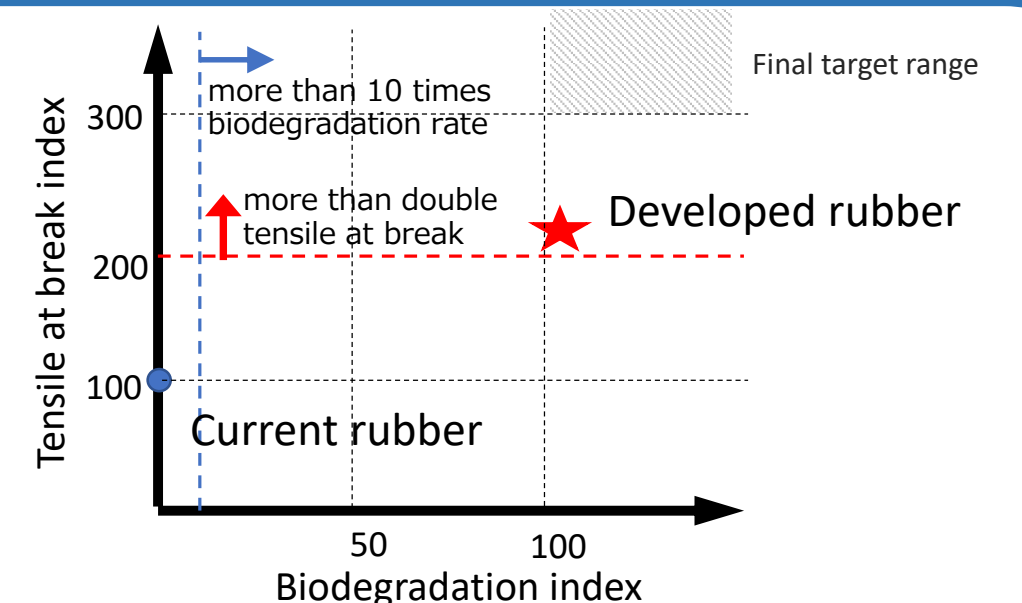


Both toughness and biodegradation are both significantly increased by introducing reversible and degradable cross-links

Summary

- We developed copolymer with biomass-derived degradable unit cooperating with academia, and achieved POC that improved degradation rate by more than 10 times faster compared to current rubber.
- We developed diene rubber copolymers with reversible/degradable cross-links cooperating with academia, and achieved more than 2 times higher strength and 10 times faster biodegradation rate.

Achievement level to target: Fully achieved the mid term target for the year 2022



No. A-11-4E

PJ: Development of Multi-lock Biopolymers Degradable in Ocean from Non-food Biomasses

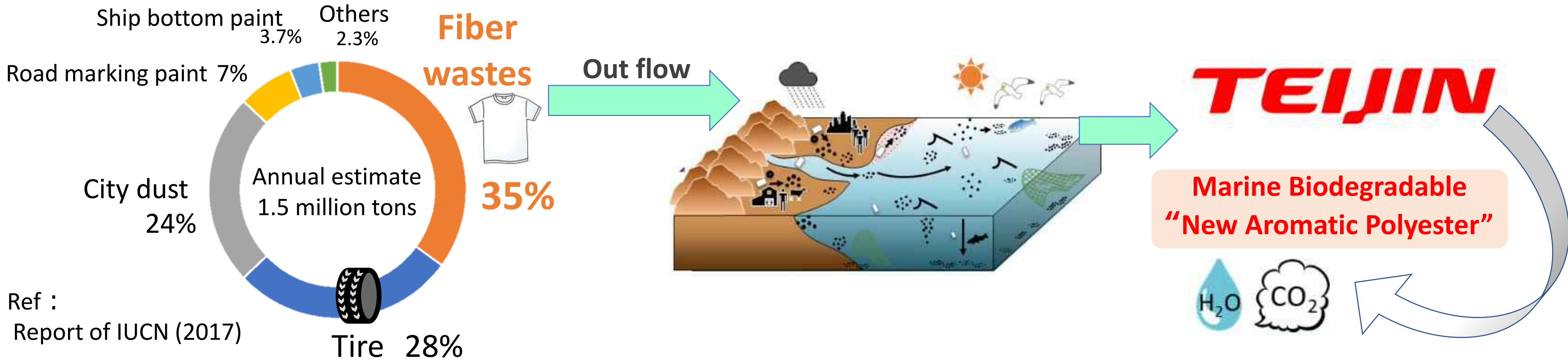
Theme: Development of highly degradable polyester-based multi-lock type bio-tough polymer and its fibers

Organization: TEIJIN LTD., TEIJIN FRONTIER Co., LTD.

Contact: TEIJIN, Material Technology Center, Tomoyoshi Yamamoto (tom.yamamoto@teijin.co.jp)



Microplastics from fibers waste



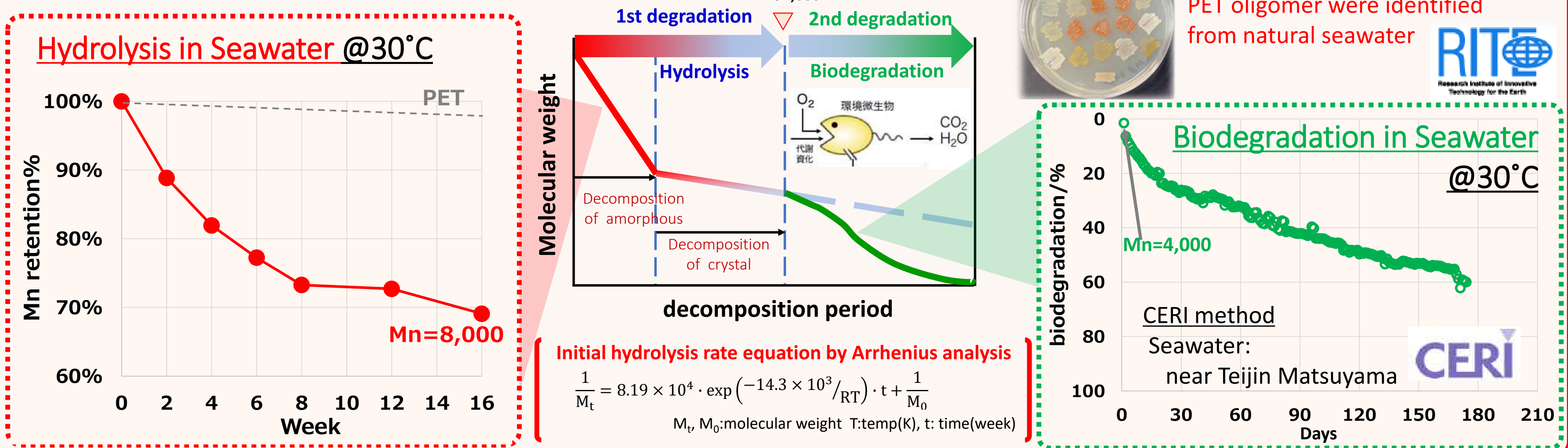
Ref : Report of IUCN (2017)

Marine Biodegradable "New Aromatic Polyester" fiber

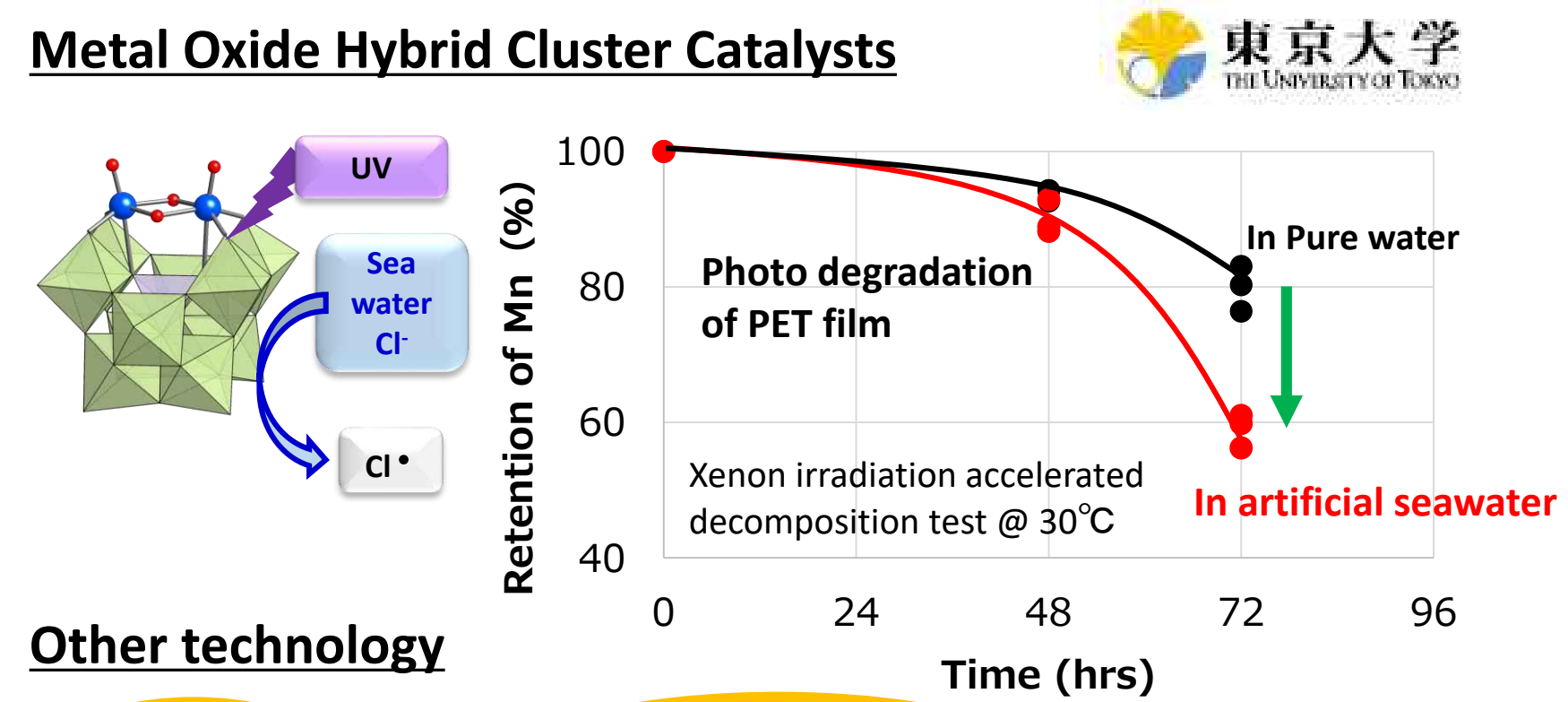
Copolymer × Decomposition Accelerator			Fiber properties Target 3cN/dtex				WAXD/SAXD				Dyeability		Washing durability	
Terephthalic acid unit	Decomposable unit	Hydrophilic unit	Denir Dtex	Filament number	Strength cN/dtex	Elongation %	Crystal -linity %	Crystal face	Crystal Size Nm	Fiber identity period (-103) nm	Long period nm	WAXD	St. after dyeing cN/dtex	St. after washing cN/dtex
			New Fiber MD1	46	24	2.9	19	68.1	110	5.5	7.56		2.4	L50 : 2.7 L100 : 2.2
			PET	50	36	4.7	36	43.2	110	3.4	5.36		3.9	L50 : 4.5 L100 : 4.5

Molecular weight (Mn)	Tg °C	Tm °C	ΔH _m J/g
1.2×10 ⁴	50	211	30

Hydrolysis & Biodegradation in Seawater



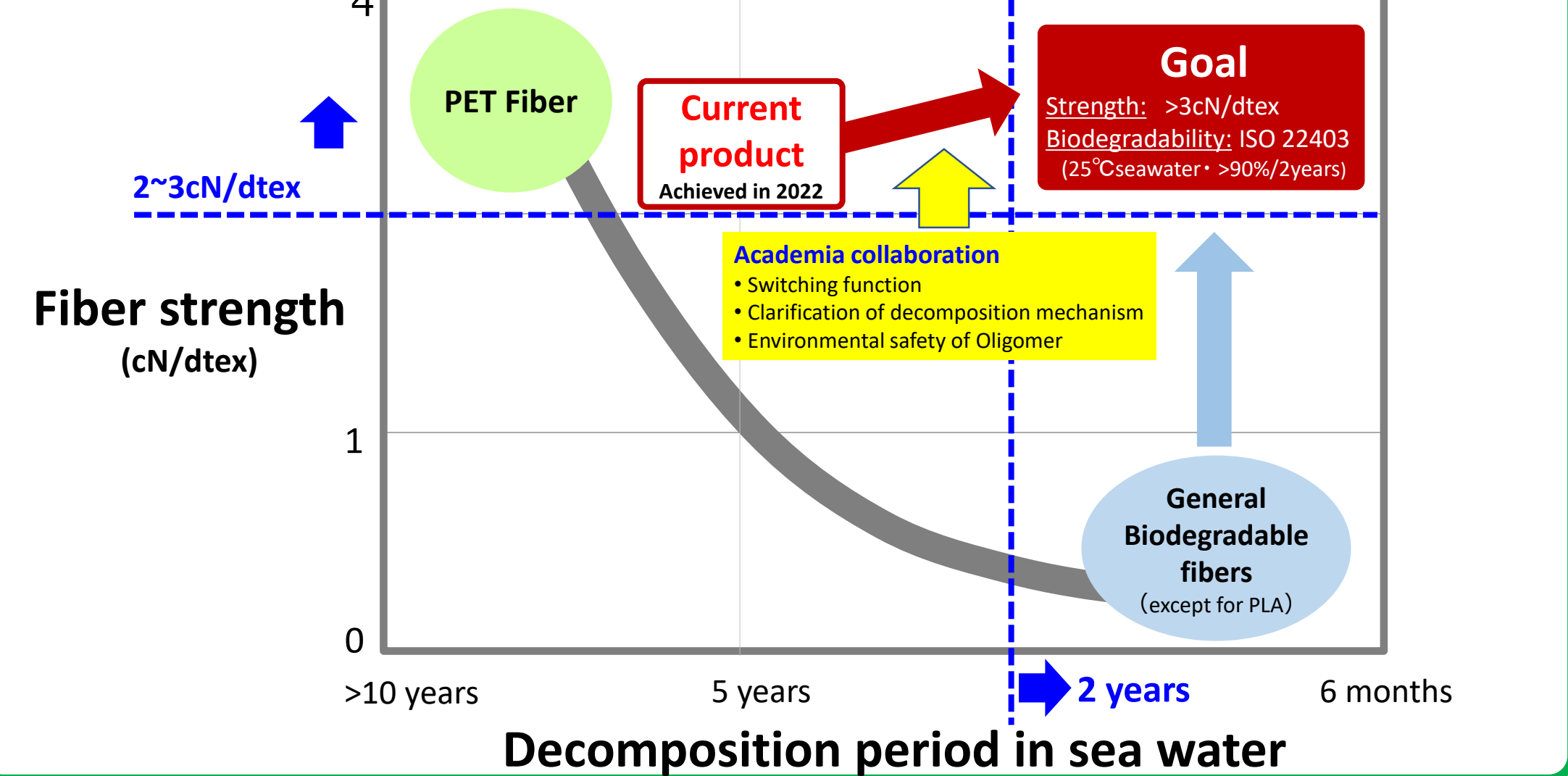
New switching function for decomposition



Other technology

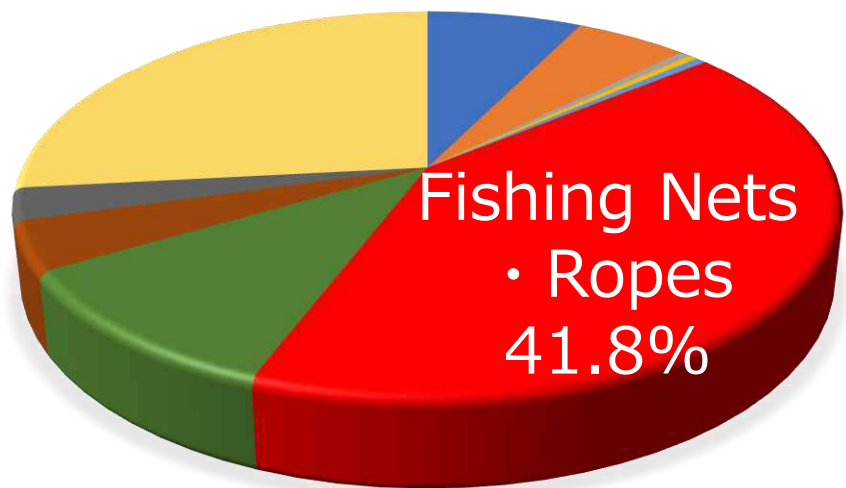
- 山形大学 (Yamagata University) - Compound technology
- 信州大学 (Shinshu University) - Development of easily degradable polymer
- 名古屋大学 (Nagoya University) - Analysis of decomposition mechanism
- 神戸大学 (Kobe University)

Next stage: Further improvement of Biodegradability



Introduction

<Classification of Drifted Plastics> <Problem of "Ghost Gear">

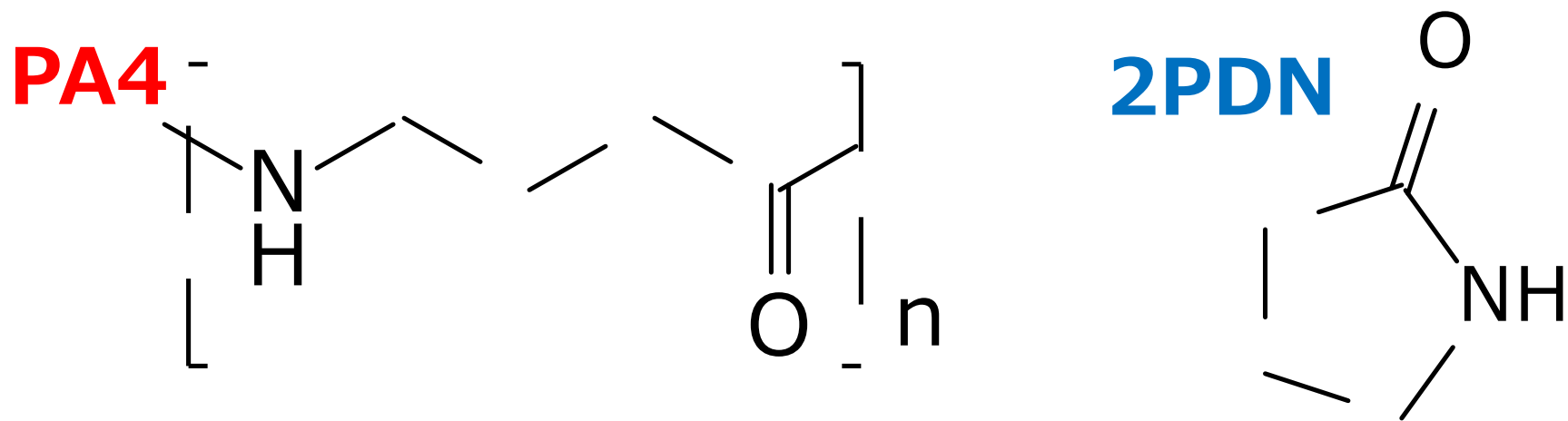


Cited from 「Recent trends in sea garbage」 by Ministry of the Environment

Cited from Wikipedia

- ✓ Fishing nets and ropes make up about 40% of the drifted plastics and they cause a problem called "Ghost Gear".
- ✓ Some biodegradable products are commercialized, however, they generally have inferior mechanical strength and degrade by hydrolysis during use.

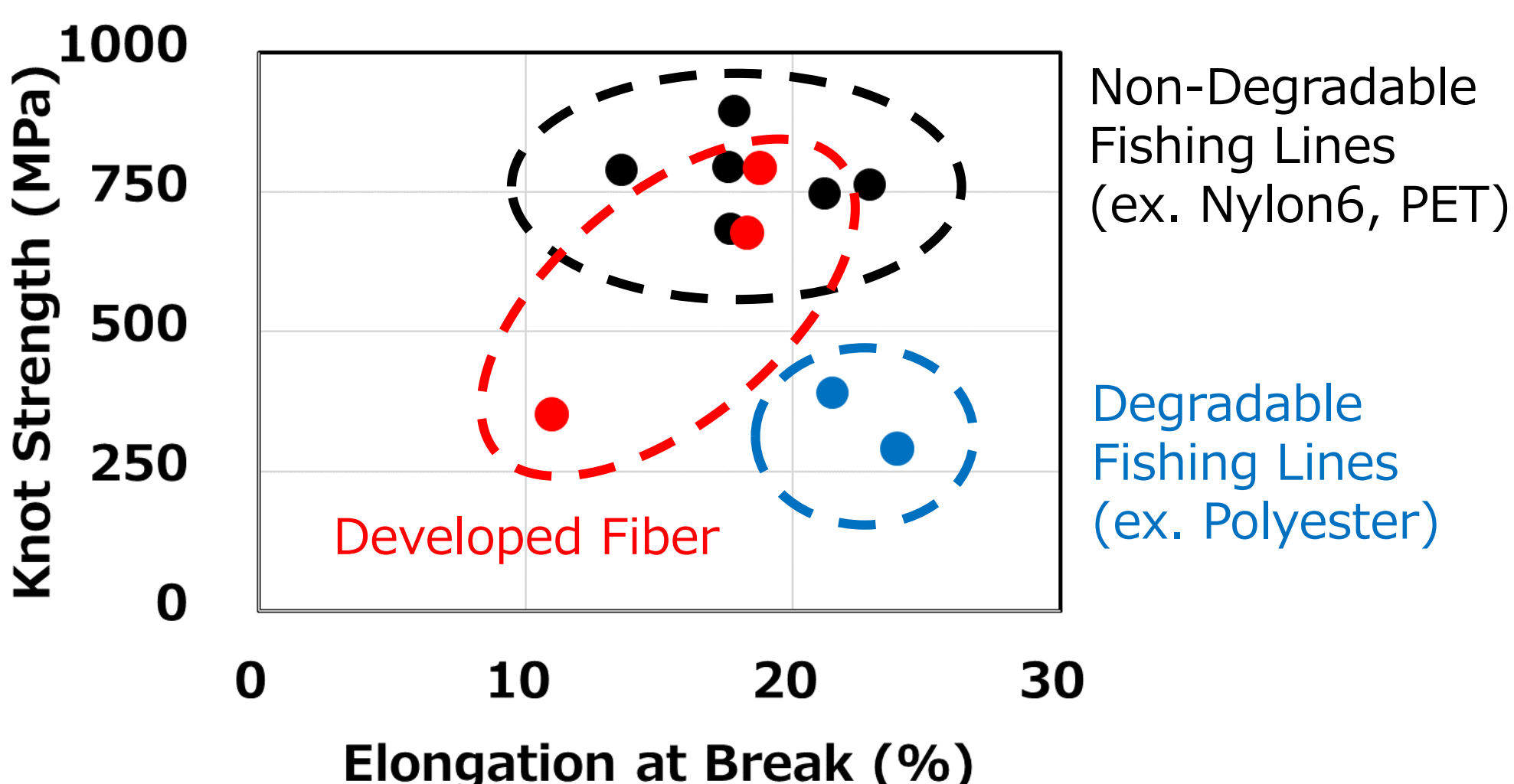
Polyamide 4 (PA4)



- ✓ PA4 is an aliphatic polyamide (nylon) synthesized from 2-pyrrolidone (2PDN).
- ✓ PA4 degrades in natural environments such as soil, sea, etc.
- ✓ In general conditions, PA4 is stable and not hydrolyzed.
- ✓ The mechanical property of PA4 is superior to Nylon6, on the other hand, PA4 has poor processability because it's thermal decomposition point is close to melting point.

Mechanical Strength of the Developed Fiber

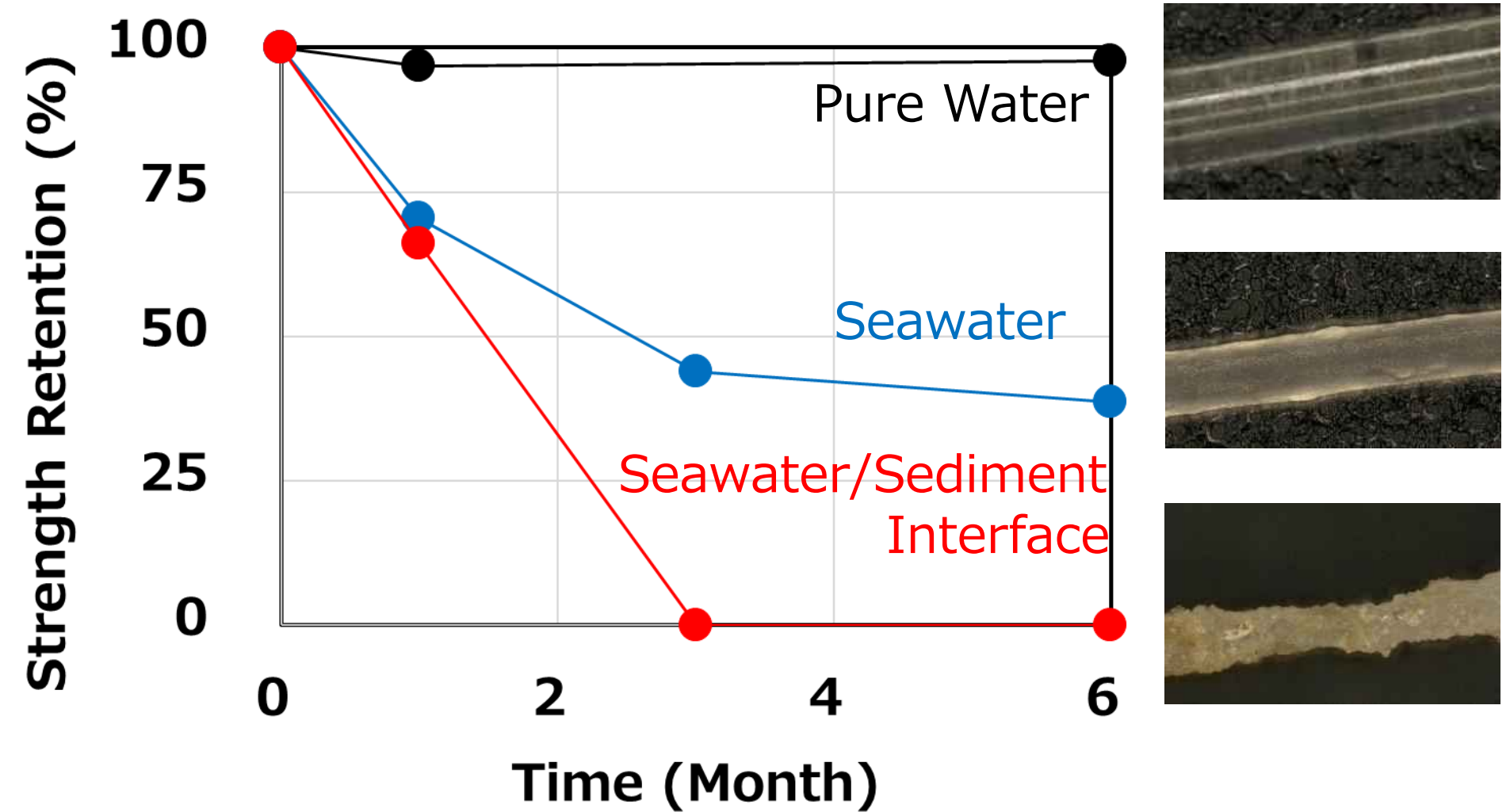
<Comparison of Knot Strength and Elongation>



- ✓ **The strength and elongation of developed fiber is equivalent to commercial non-degradable fishing lines.**

Marine Biodegradability

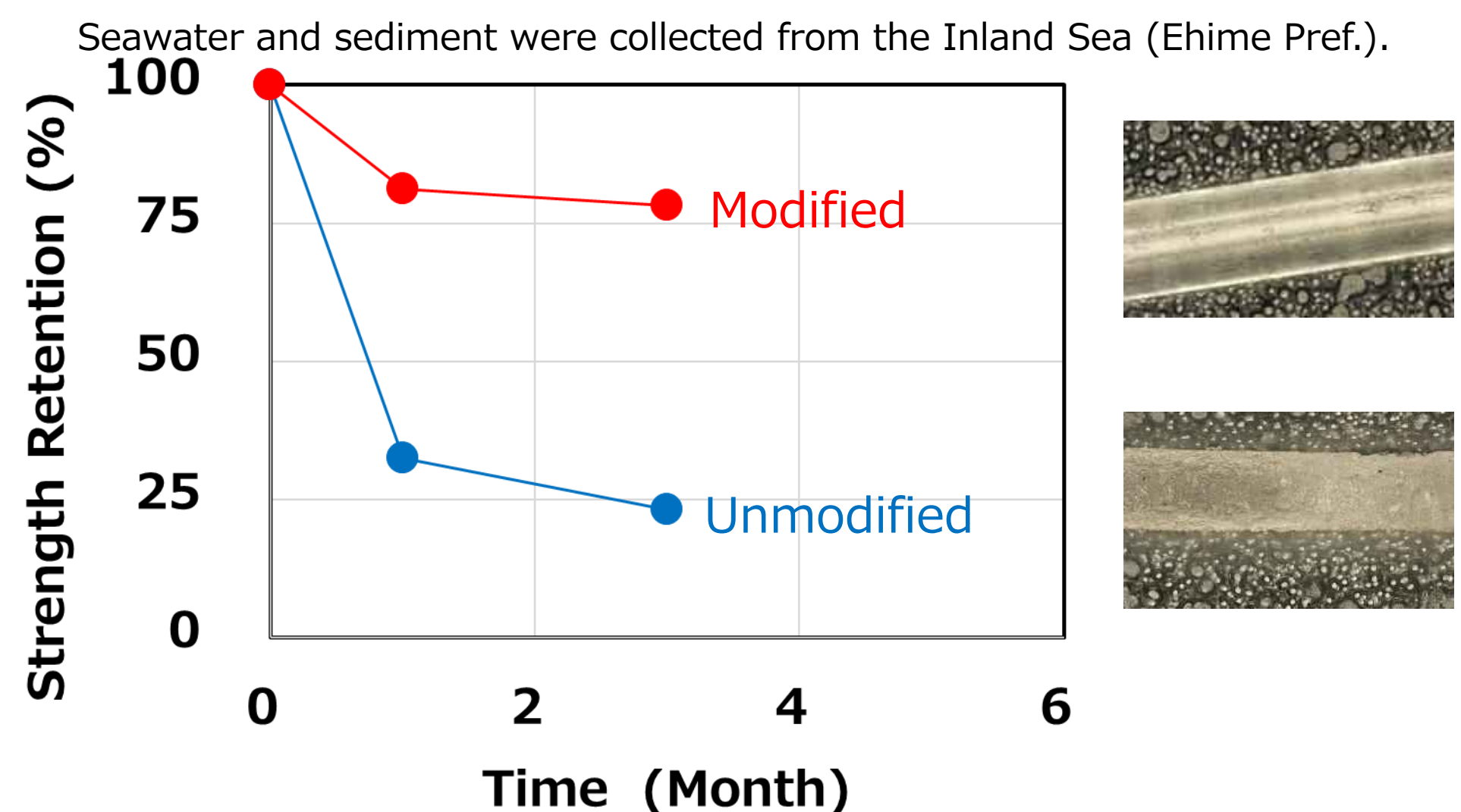
<Degradation Test of the Developed Fiber in Laboratory>
Seawater and sediment were collected from the Pacific Ocean (Fukushima pref.).



- ✓ In pure water, there was almost no reduction in strength.
- ✓ At seawater/sediment interface, degradation was faster than in seawater.

Control of Degradation

<Comparison of Degradation at Seawater/Sediment Interface>



- ✓ **Degradation rate of modified fiber was reduced to 1/3.**

Summary

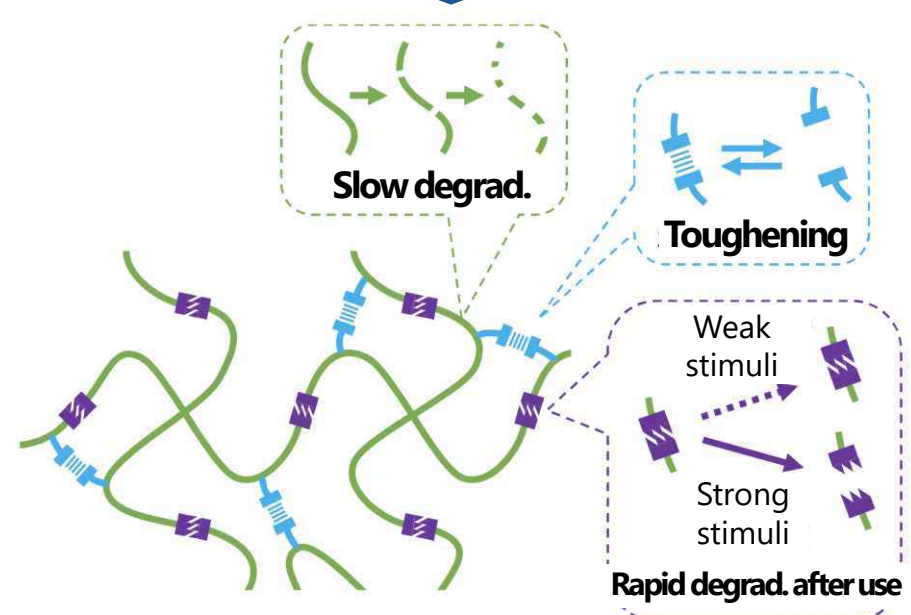
- ✓ We are developing biodegradable and tough biopolymer for fishing gears based on polyamide 4.
- ✓ The strength and elongation of developed fiber is equivalent to commercial non-degradable fishing lines.
- ✓ The degradation rate of modified fiber was reduced to 1/3.

Ongoing / Future Plan

- ✓ Collect degradation data under various environments
- ✓ Develop utilization technique of biobased raw materials
- ✓ Combine with other degradation control technologies (ex. Sheath-core conjugate fiber)

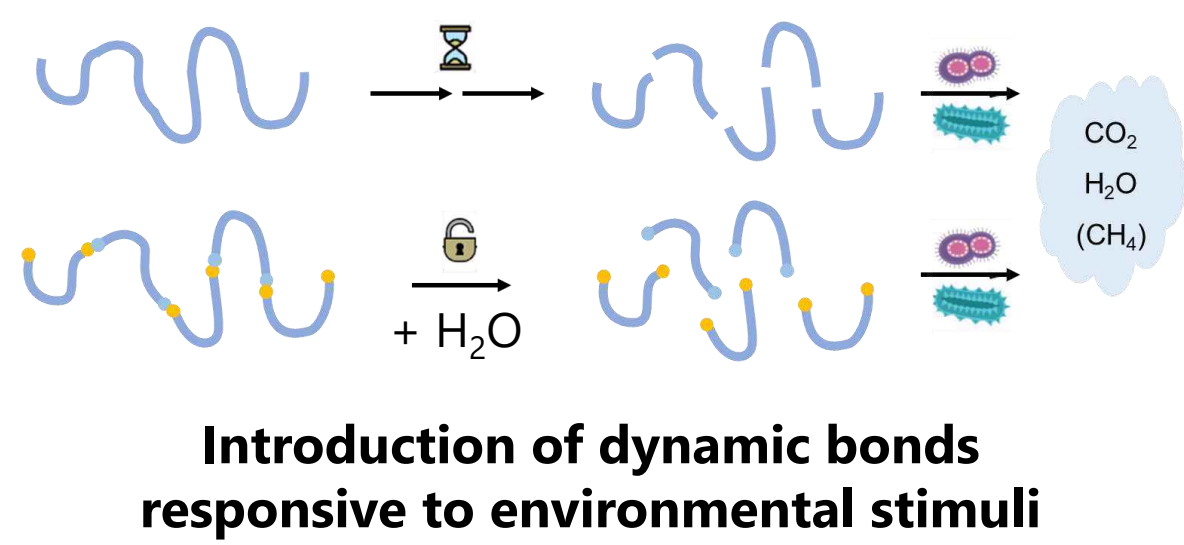
Introduction

There is generally a trade-off between degradability and mechanical strength.



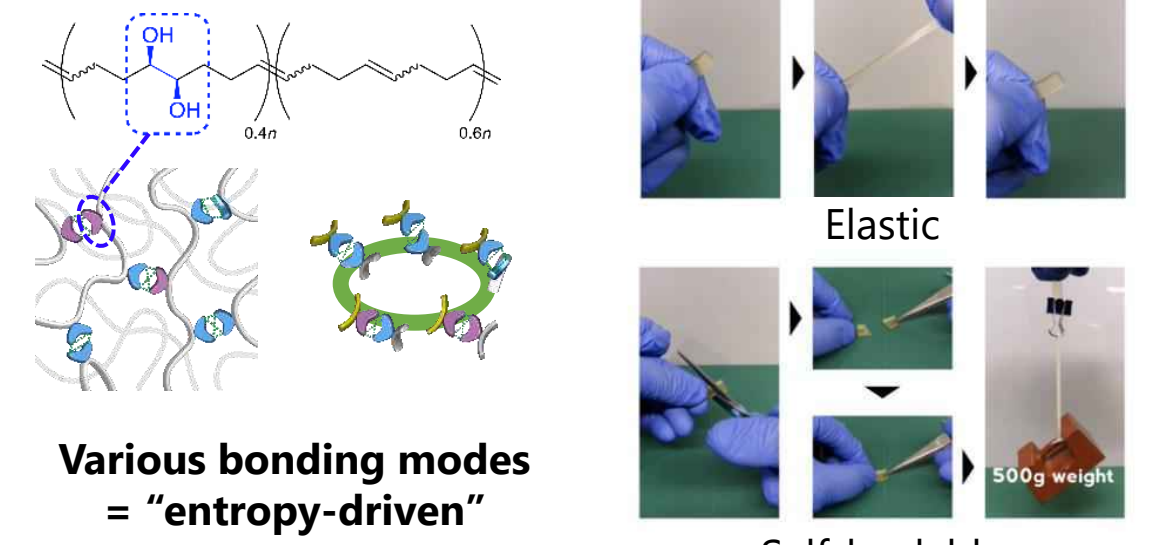
Need for high mechanical strength during use and rapid degradation in environment after use

Topic 1: Development of multi-lock mechanism

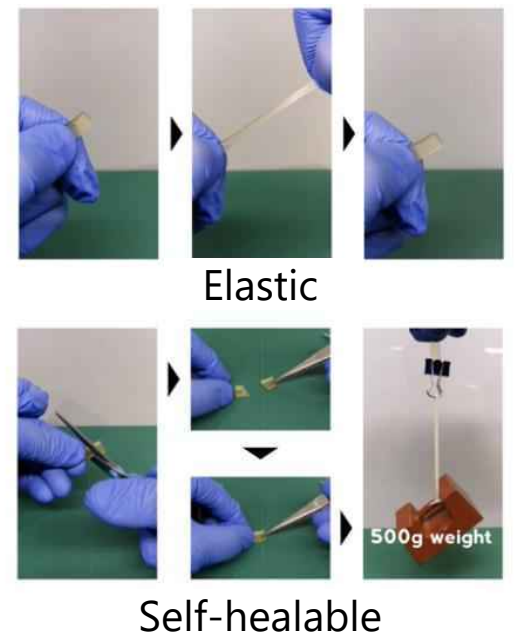


Accelerate degradation by facilitating the time-consuming oligomerization process

Topic 2: Combine crosslinking and degradability

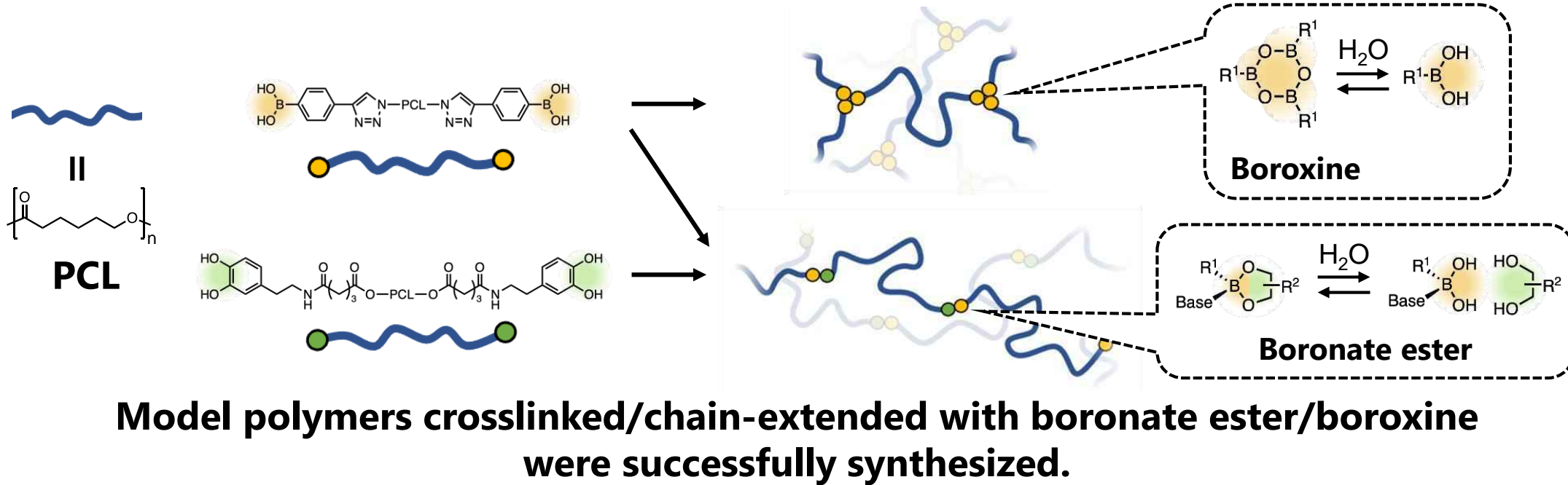


Various bonding modes = "entropy-driven"



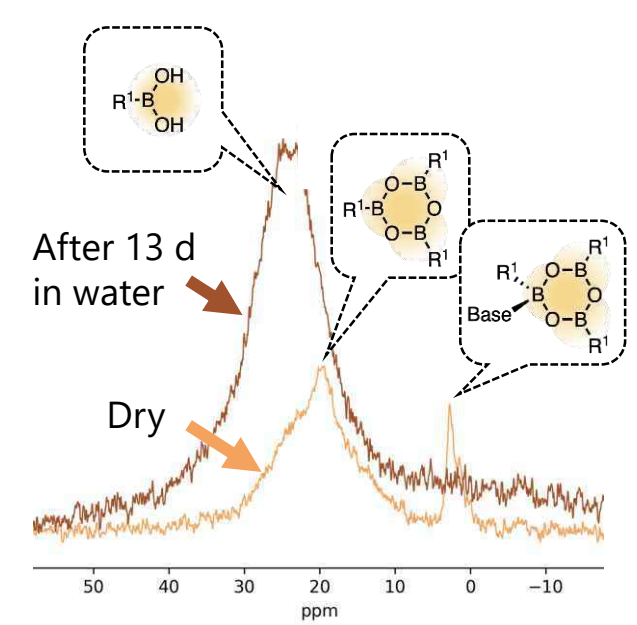
Realize excellent toughness using the "entropy-driven" multidentate hydrogen bonds

Tuning degradability by dynamic B chemistries



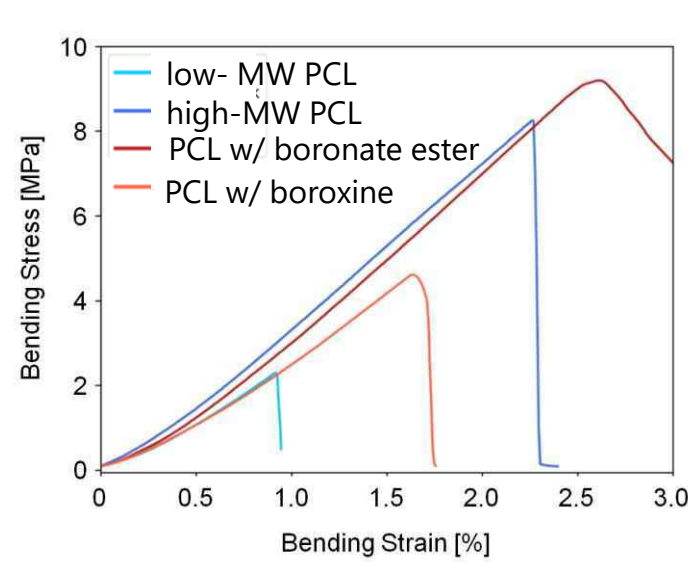
Model polymers crosslinked/chain-extended with boronate ester/boroxine were successfully synthesized.

Solid-state ¹¹B NMR



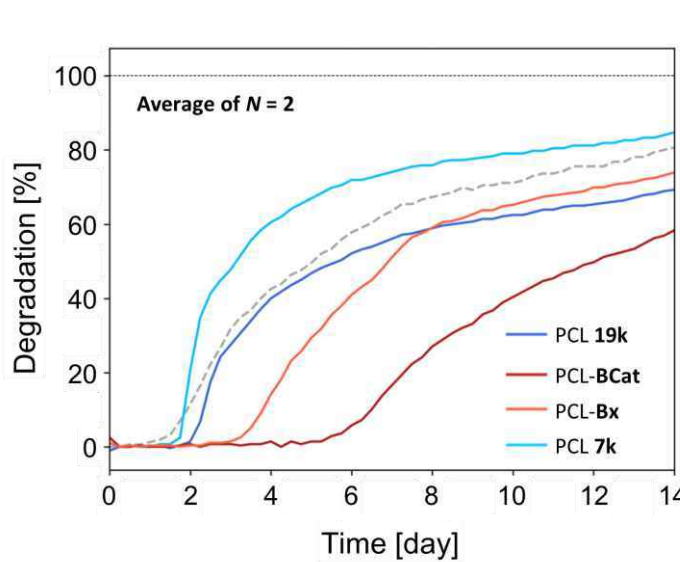
Bond dissociation by water was confirmed.

3-point bending test



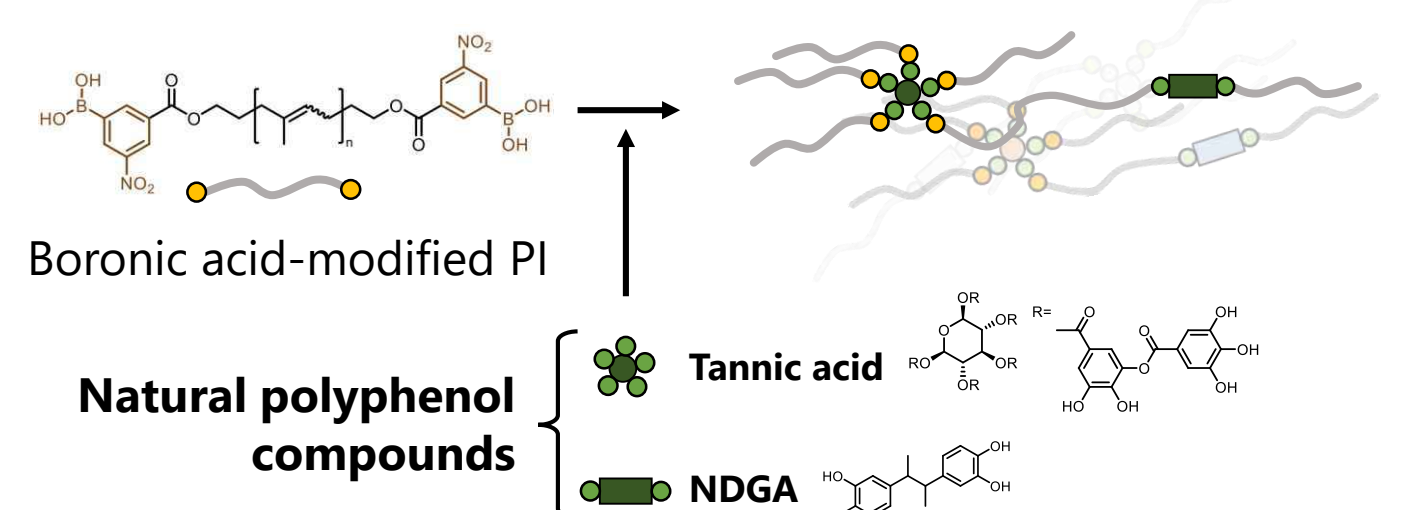
Mechanical performance was enhanced without sacrificing biodegradability.

Marine biodegradability (BOD)



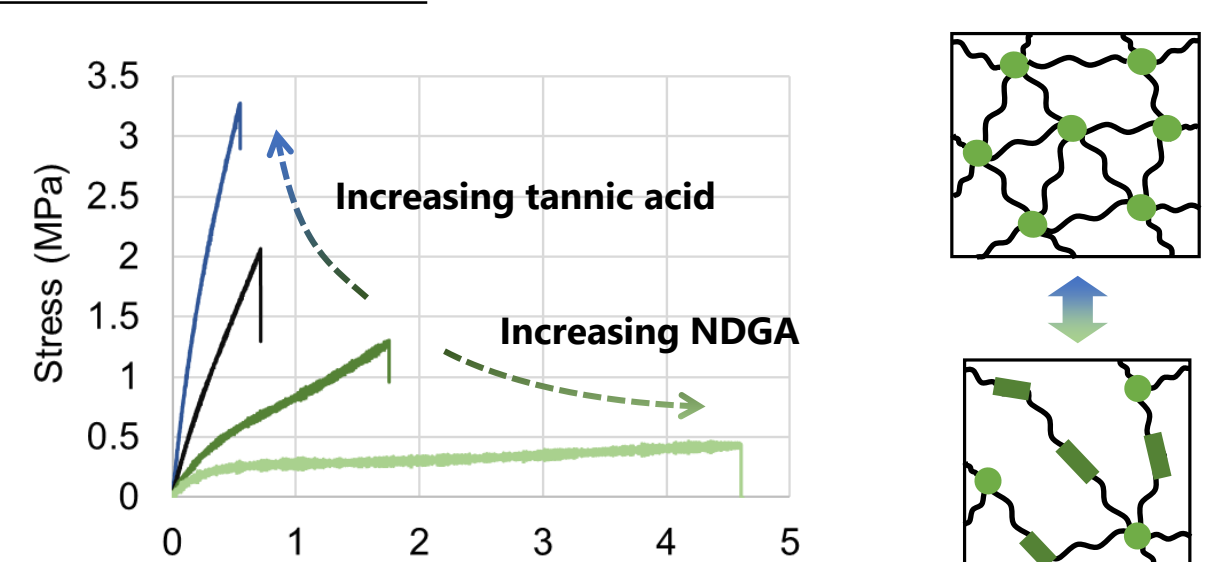
Rubber crosslinked by boronate esters

We developed polyisoprene (PI) rubber crosslinked by boronate ester.



Natural polyphenol compounds

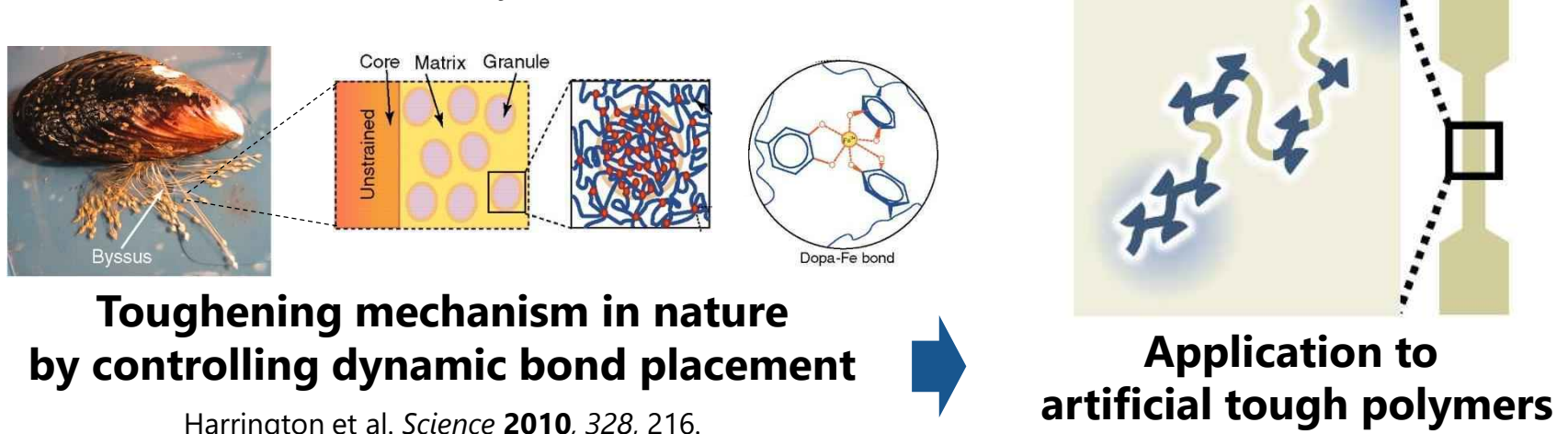
Uniaxial tensile test



Rubbery materials were obtained solely by boronate ester crosslinking.

Toughening by controlled placement of dynamic bonds

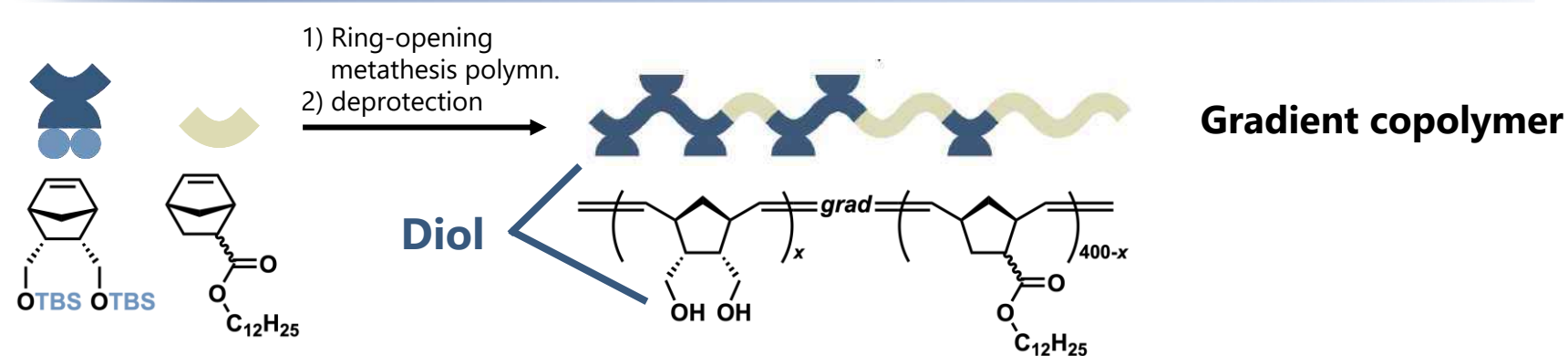
Publication: Ishizaka et al. *Polymer* 2022, 246, 124748.



Toughening mechanism in nature by controlling dynamic bond placement

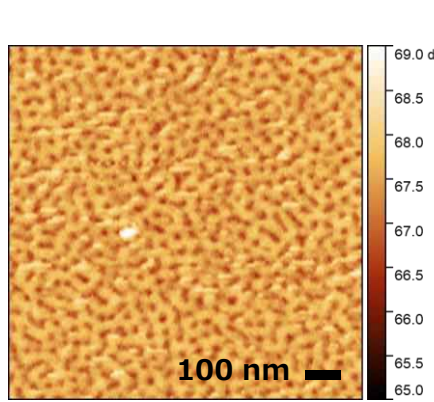
Harrington et al. *Science* 2010, 328, 216.

Application to artificial tough polymers



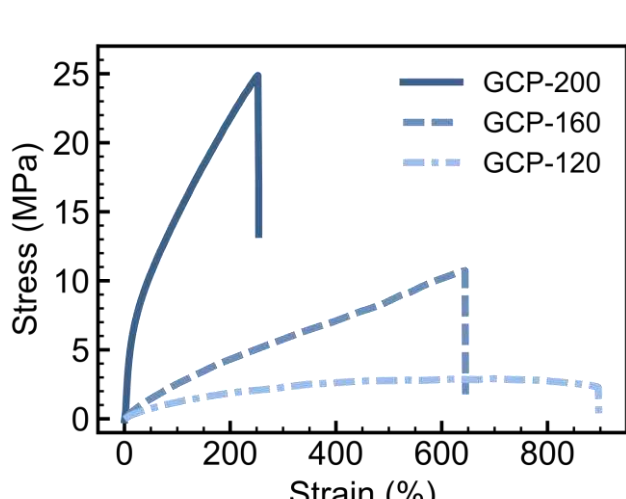
Gradient copolymer

Atomic force microscopy

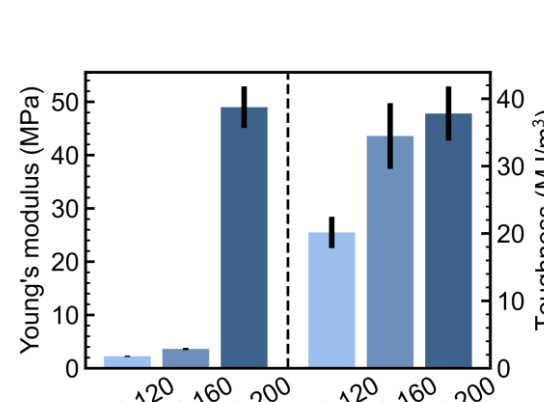


Nano phase separation

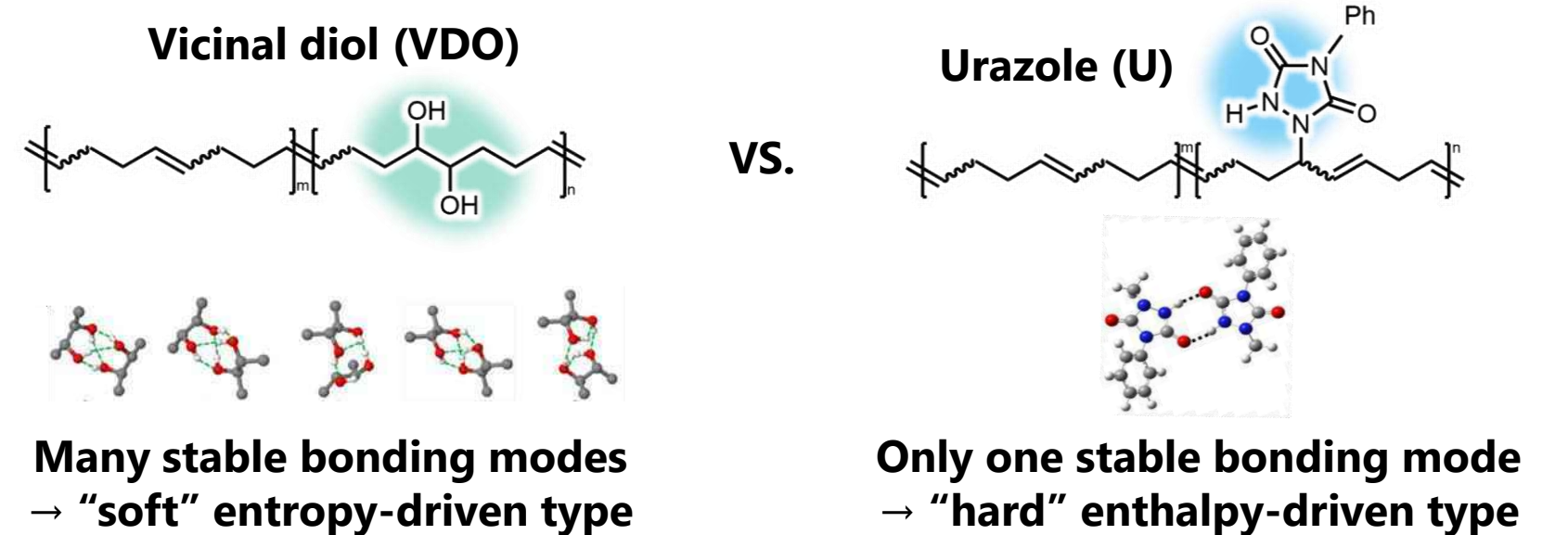
Uniaxial tensile test



Toughness was realized by inhomogeneous distribution of hydrogen-bonding diols

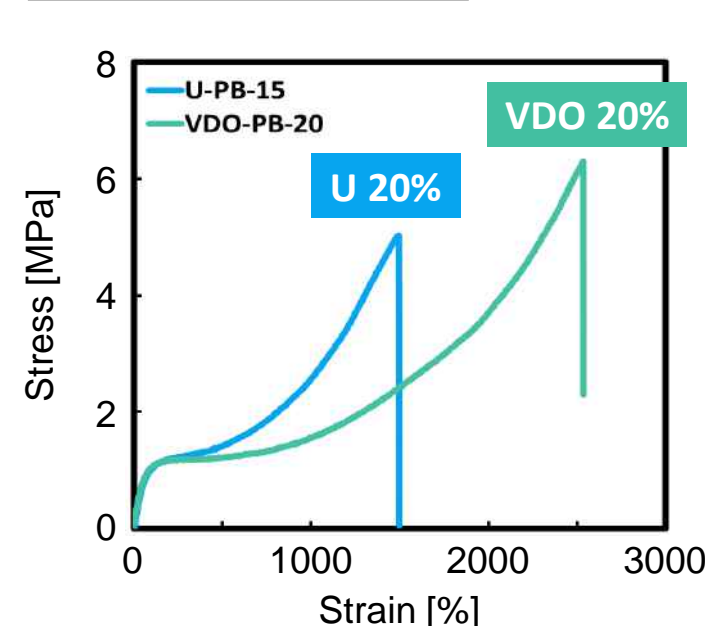


Evaluation of entropy-driven dynamic bonds

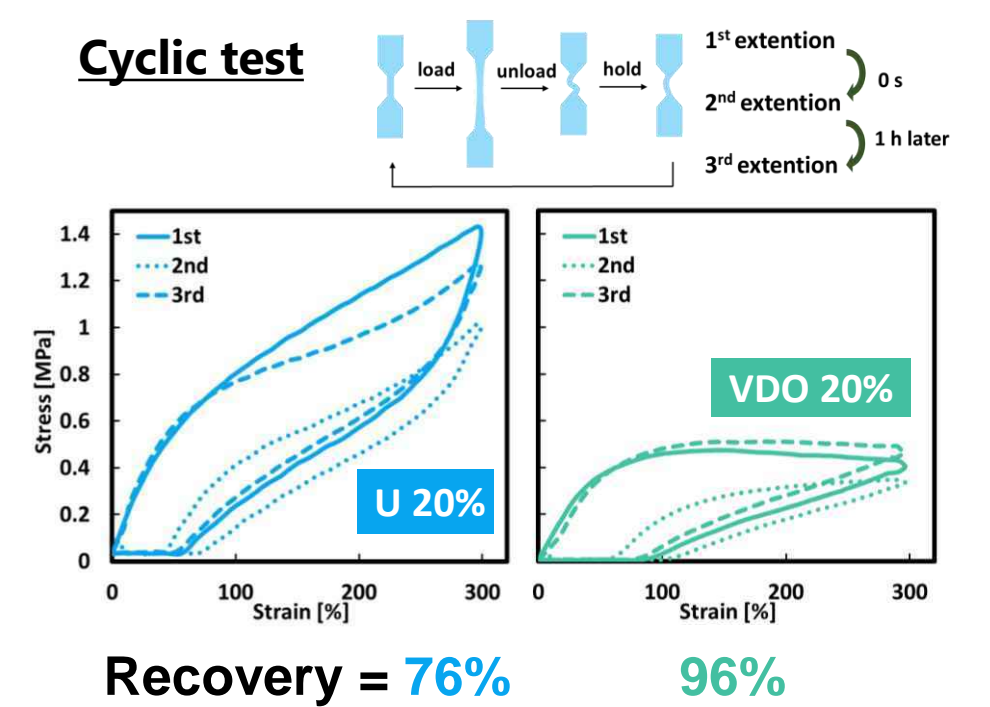


Evaluate the characteristics of the bonds by examining the mechanical properties of the model polymers

Uniaxial tensile test



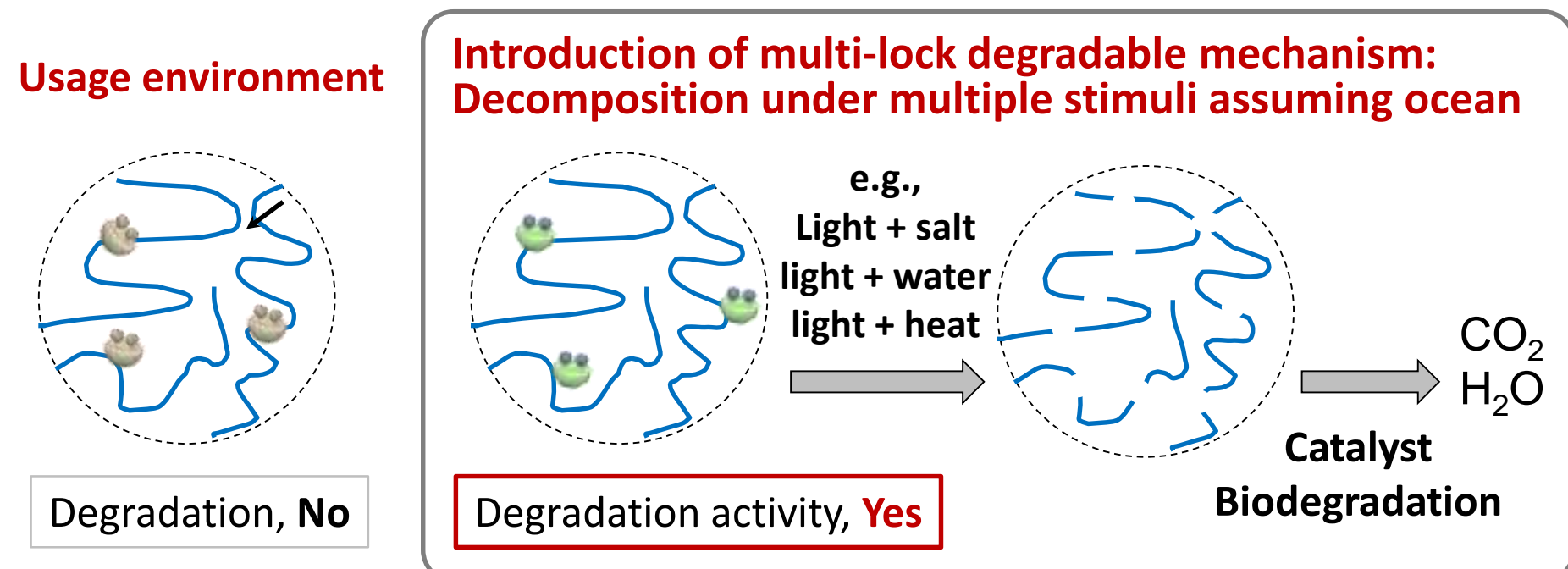
Cyclic test



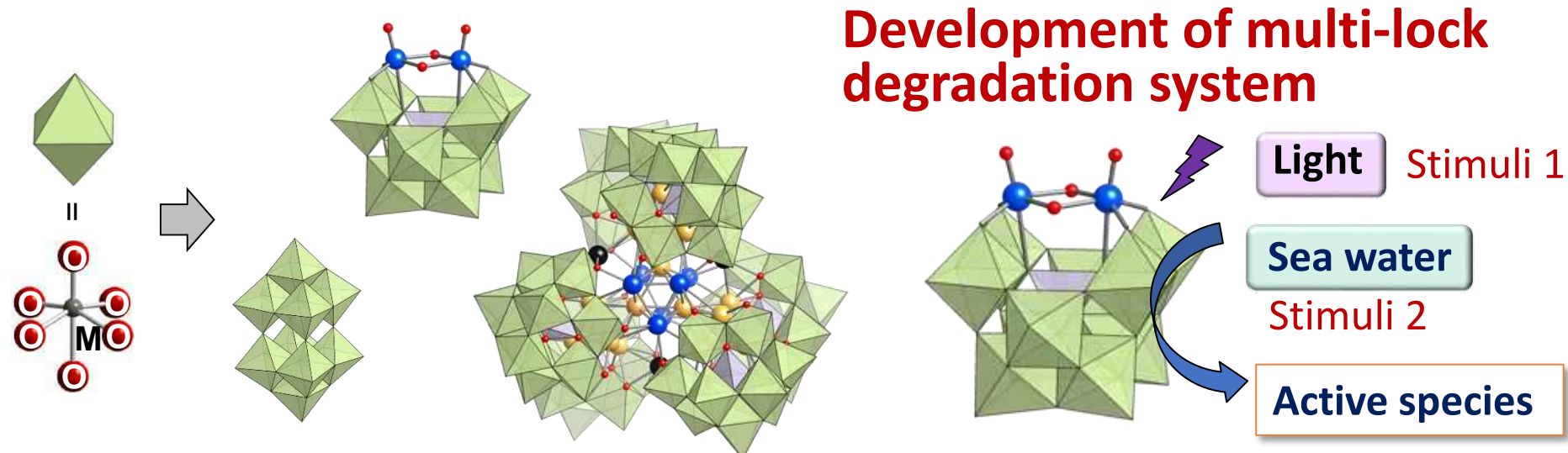
Physical crosslinks between VDOs are weaker but easier to reform → unique feature of soft, entropy-driven multidentate hydrogen bonds

1. Research contents in this project

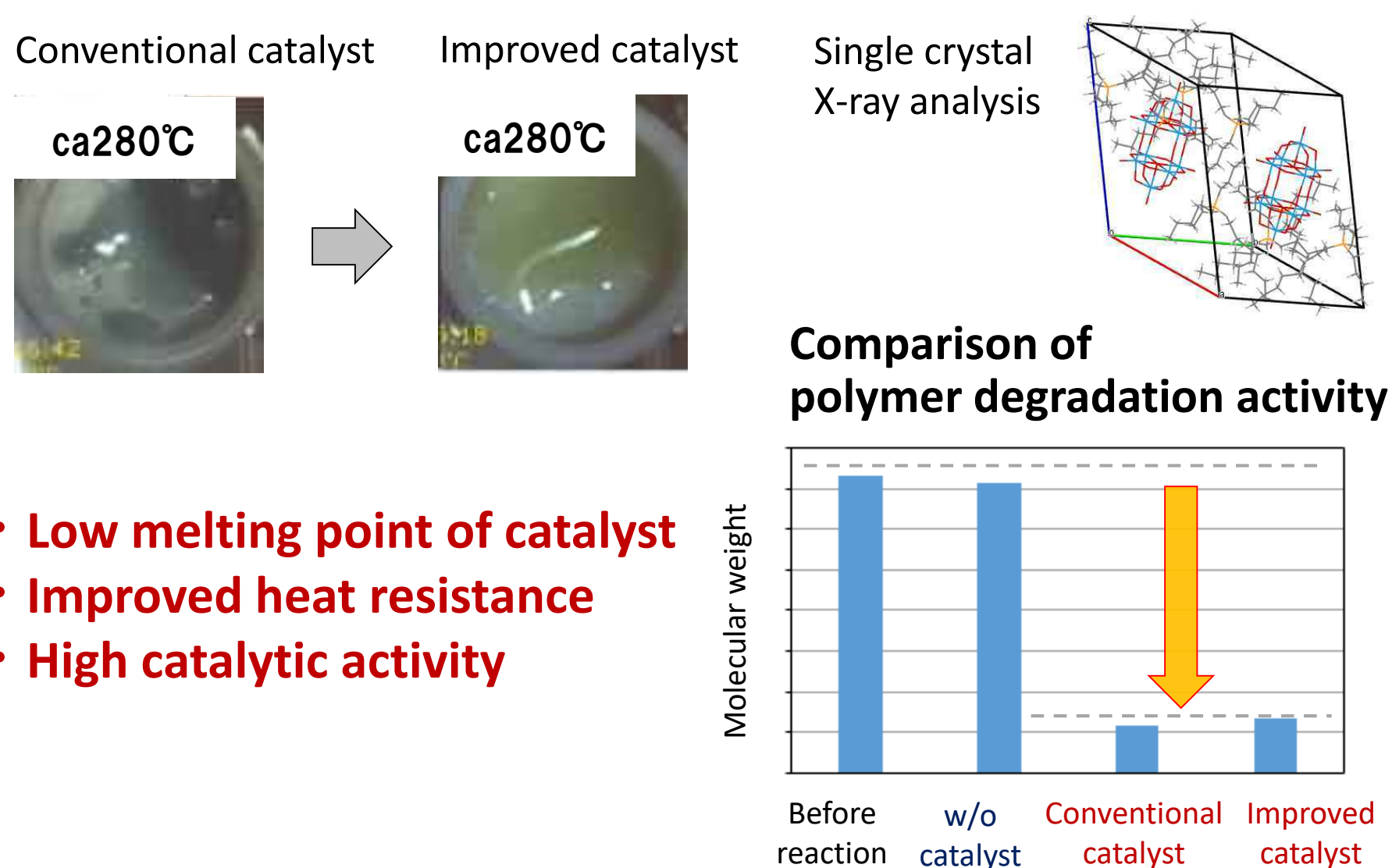
The purpose of this study is to develop polymers that have both multi-lock degradation mechanism and toughness in ocean. By utilizing our synthesis methods of metal oxide clusters, we will develop catalysts that exhibit degradation activity when two or more of the stimuli, such as light, heat, oxygen, water, salts, enzymes, microorganisms, are present at the same time, and use these catalysts to perform on-demand degradation of polymers.



Design of metal oxide cluster catalysts

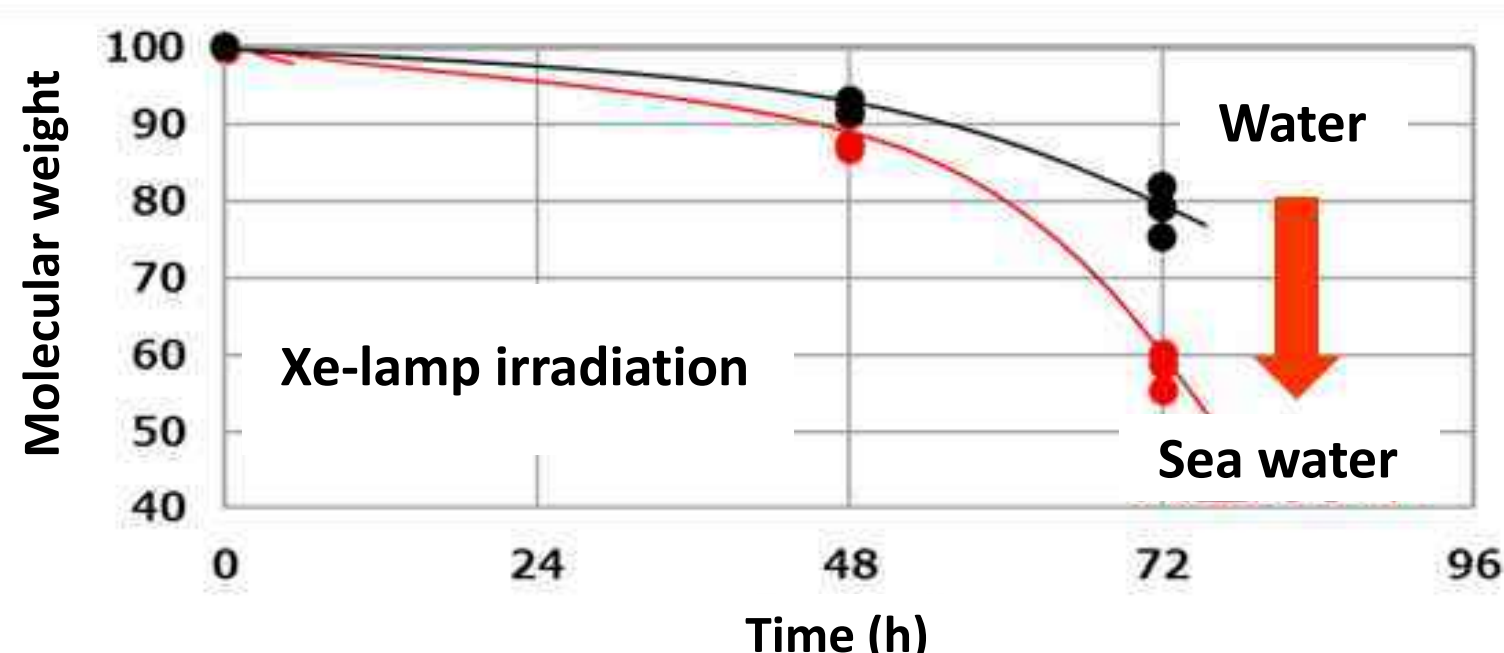


Development of catalyst with heat resistance



3. Collaboration with companies

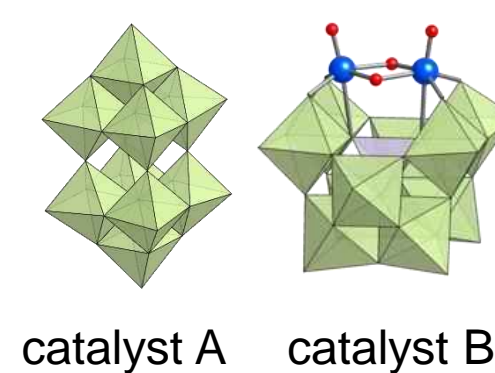
Degradation of catalyst-containing PET polymer



Catalysts accelerate photodegradation of PET polymer in artificial seawater

2. Common Issues

Preparation of catalyst-containing polymer films

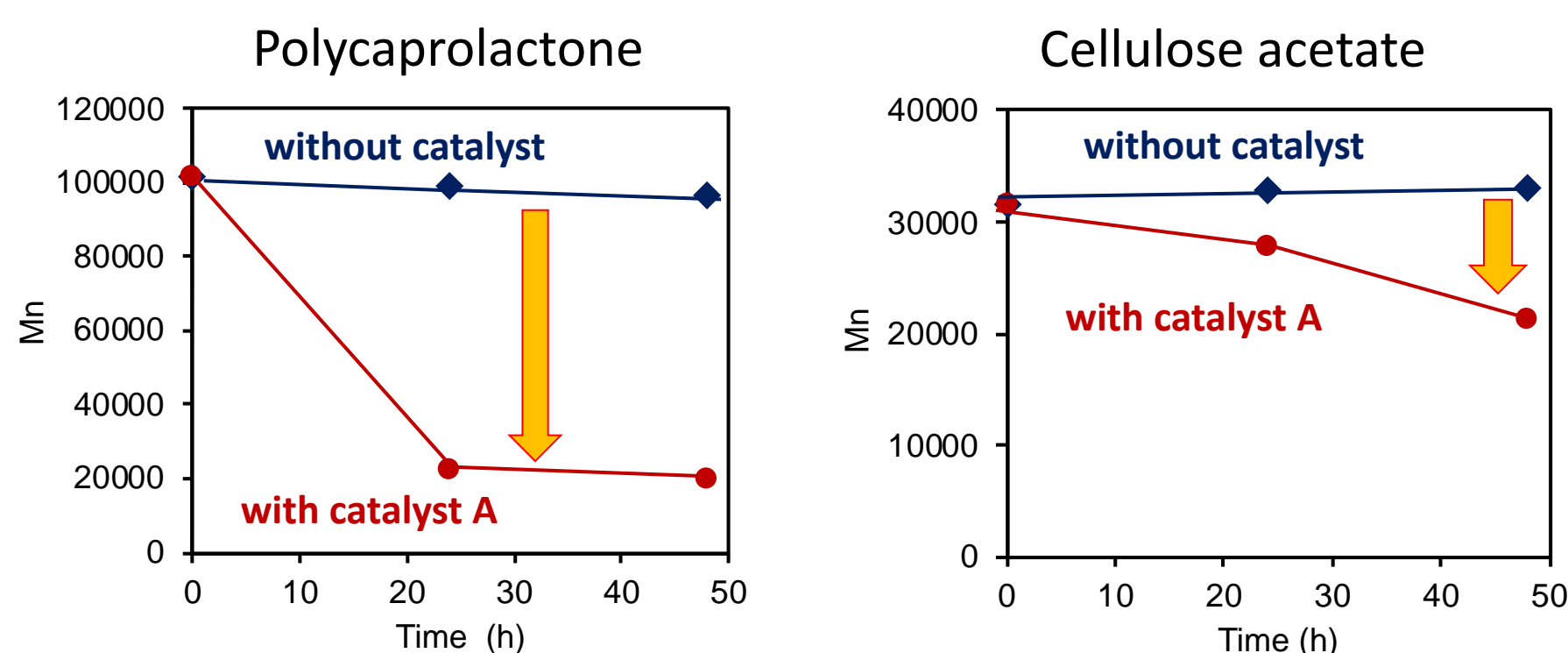


Sample	M_n
Polycaprolactone	112458
Polycaprolactone + cat A	106664
Polycaprolactone + cat B	115858

Catalysts do not affect molecular weight of polymers during film preparation

Photodegradation of catalyst-containing polymer film

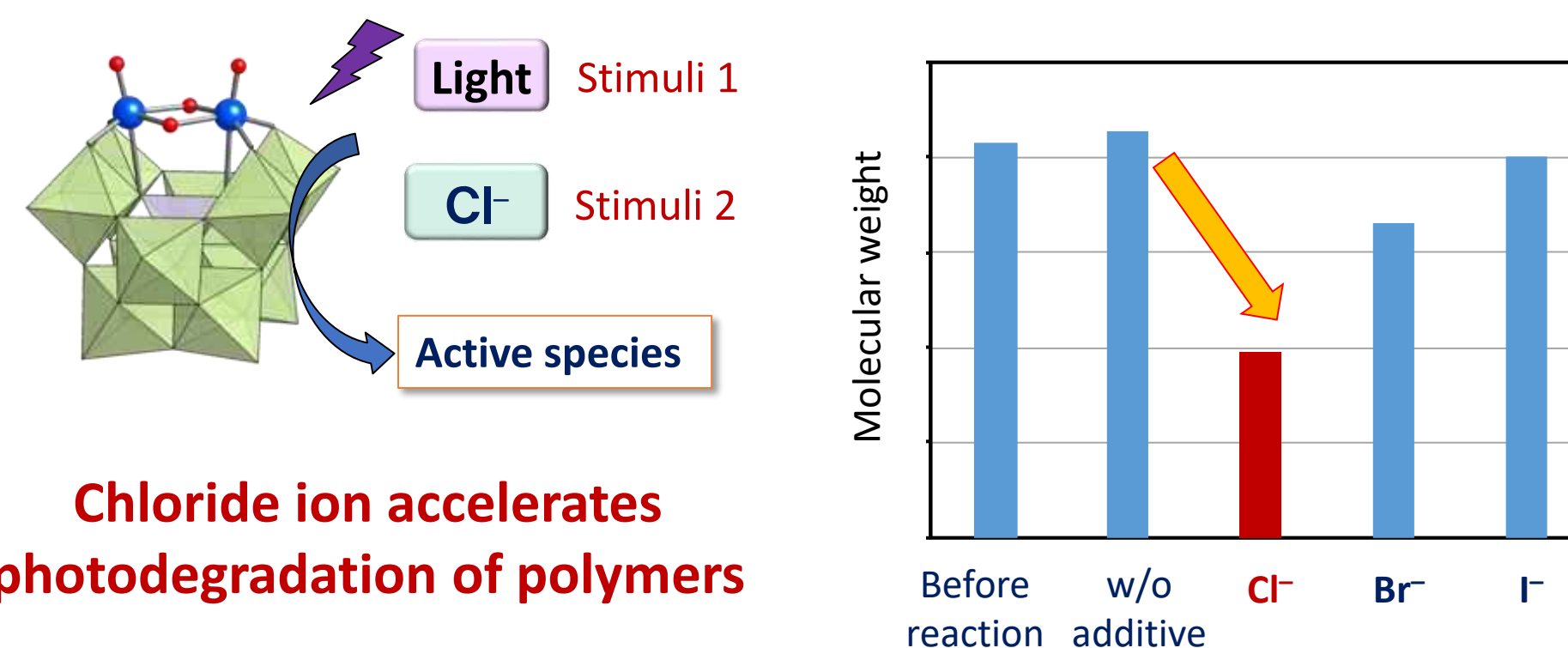
Molecular weight change of polymers under Xe-lamp irradiation



Catalyst-containing polymer films undergo accelerated photodegradation

Multi-lock degradation using chloride ions (Cl⁻)

Additive effect on polymer degradation under Xe-lamp Irradiation



4. Collaboration with academia

University of Tokyo

Yamagata University

5. Future Plans

Computational chemistry for elucidation of reaction mechanisms

Consideration for reducing the amount of catalyst and additives

Development of new multi-lock degradation mechanisms

Investigation combining catalytic degradation and biodegradation