

Innovative Circular Technologies for Harmful Nitrogen Compounds/ To Solve Planetary Boundary Issues

Theme 2. Recycling nitrogen compounds in wastewater to ammonia resource Theme 2-2. R&D on ammonia recycling by separation and concentration R&D of NH₄⁺ adsorbent and recycling processes

Presenter : Tohru Kawamoto (National Institute of Advanced Industrial Science and Technology (AIST)) PM : Dr. Dr. KAWAMOTO Tohru , National Institute of Advanced Industrial Science and Technology (AIST) Implementing organizations : National Institute of Advanced Industrial Science and Technology (AIST), The University of Tokyo, Waseda University, Tokyo University of Agriculture and Technology, Kobe University, Osaka University, Yamaguchi University, Kyowa,Hakko Bio Co., Ltd.,

ASTOM Corporation, Toyobo Co., Ltd., FUSO Corporation, Ube Industries, Ltd,

Position in the Project





Target of Theme 2 for FY2029 : demonstration with a pilot plant with and enrichment on a scale of $5\sim$ 15 m³/d.

Position of AIST:R&D of Adsorbents and Adsorption Technology for Selective NH₄⁺ Enrichment

Target of AIST for FY2029:Establishment of materials and basic processes for the pilot demonstration of NH_4^+ adsorption and recovery.

Details & Items of R&D



R&D of selective ammonium adsorbent its utilization processes for the enrichment of the ammonium for the recovery as a resource



R&D Items

- Research and development of adsorbents that can selectively recover and concentrate NH₄⁺ ions in water
- Basic design of a system to produce ammonia resources by using the adsorbents

Achievement(1) Ion-exchange adsorption / desorption(AIST)

- Adsorbent with selective adsorption NH₄⁺ by exchange with K⁺
- NH₄Cl solution is generated by passing KCl solution, enabling separation from coexisting substances.



Copper-substituted Prussian blue-type complexes with Optimized composition

Achievement(1) Ion-exchange adsorption / desorption(AIST)



- Adsorbent that can selectively adsorb NH₄⁺ even from wastewater
- 400 times recyclability, washing is considered for more improvement

Dependence of the amount of NH_4^+

on number of cycles (column test)

Adsorption isotherms for NH₄⁺ ions from pure water NH₄HCO₃ solution and from sewage sludge digestate





 >60 MOFs synthesized and screened for NH₄⁺ adsorption out of Reviewed 240 MOFs reviewed

Scrooped MOEs (nartially)

• Several MOF candidates discovered (First time for MOFs)

Screening for resistant MOFs Water-resistance examined ...ca. 240 Water-resistant MOFs ...ca. 120 Affordable MOFs ...47 ╋ 64 Other candidates ...15

<u>Screence wors (partially)</u>				
MIL-53(Cr)	AITCS-2	PCN-333(AI)	UiO-67-BIPY	MFM-303(AI)
MIL-101(Cr)	DUT-5(AI)	PCN-333(Fe)	HKUST-1	Ni ₃ (BTP) ₂
MIL-53(AI)	HP-DUT-5(AI)	PCN-777(Zr)	H ₂ Q-HK	NENU-3
MIL-53-NH ₂ (AI)	DUT-67(Zr)	SUMOF-7	Zn ₃ IPA ₃	NENU-5
HP-MIL-53(AI)	DUT-68(Zr)	LnBTN	kag-MOF-1	NENU-11
HP-MIL-53-NH ₂ (AI))DUT-69(Zr)	Eu ₆ (OH) ₈ (NDC) ₆	Cu ₆ (Trz) ₁₀	AmNaCr
MIL-68(In)	MOF-467	UiO-66	MOS-1	$NH_4[Zn(HCO_2)_3]$
MIL-68-NH ₂ (In)	MOF-808-Formate	UiO-66-NH ₂	MOS-2	LOF
MIL-125-NH ₂ (Ti)	MOF-808-OAc	UiO-67	MOS-3	K-adp-Zn(ox)
CAU-1	PCN-222	HP-UiO-66	IISERP-MOF2	$ZnBP-NH_4$
AUBM-1	PCN-223	HP-UIO-66-NH ₂	Mn-MOF-74	ZrP
AITCS-1	PCN-225	HP-UiO-67	Mg-MOF-74	ZrPOF-EA
[Cu ₂ (BTC) ₂ (H ₂ O)))] (Me))		(NMe.),H-f(Cu.Cl)	(BTC)-)]-





Position in the project

R&D of Adsorbents and Adsorption Technology for Selective NH₄⁺ Enrichment

Target for FY2029

Establishment of materials and basic processes for the pilot demonstration of NH_4^+ adsorption and recovery.

R&D items

- Research and development of adsorbents that can selectively recover and concentrate NH₄⁺ ions in water
- Basic design of a system to produce ammonia resources by using the adsorbents

Achievement 1(AIST)

- Adsorbent that can selectively adsorb NH₄⁺ even from wastewater
- 400 times recyclability, washing is considered for more improvement

Achievement 2(Univ. Tokyo)

 Several MOF candidates discovered from 60 samples(First time for MOFs)

